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In-Situ Manufacturing of SiC-Doped MgB2 Used for Superconducting Wire

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https://doi.org/10.18280/acsm.450210	ABSTRACT
Received: 1 June 2020	Magnesium diboride (MgB2) is a highly potential superconducting material, in
Accepted: 30 March 2021	substitution of Nb ₃ Sn, which has a critical temperature of ~ 39 K. This synthesis and
Keywords: MgB ₂ , silicon carbide, superconductor, heat treatment, SS304, critical temperature	manufacturing of MgB ₂ wire were conducted by <i>in-situ</i> powder in tube (PIT). The method doped with silicon carbide (SiC) was aimed to study the effect of phase formation on carbon substitution and morphological characteristics with the motivation to improve superconductivity properties. Magnesium, boron, and SiC powders were synthesized and functionally processed with stainless Steel 304 tube. Heat treatment was conducted at $750000 = 105000$ for 21 m size of the state of the st

superconductivity properties. Magnesium, boron, and SiC powders were synthesized and functionally processed with stainless Steel 304 tube. Heat treatment was conducted at 750°C, 800°C, and 850°C for 2 hours followed by furnace cooling. Characterization was carried out by x-ray diffractometer (XRD), scanning electron microscopy (SEM), and cryogenic magnet testing. The results showed that 1% SiC optimally increased the zero critical temperature of MgB₂ ~ 37.18 K along with the sintering at 750°C for 2 hours.

1. INTRODUCTION

Superconductor has a great potential to replace the current conducting wire with the advantage of efficient electrical conductivity, with the support of many studies on the synthesis and characterization of magnesium diboride initiated by Nagamatsu et al. with the achievement of zero resistivity at a critical temperature of 39 K [1-3] Many researchers have carried out sustainable research by modifying out carbide-based dopants to manufacture downstream superconducting wire products for industrial applications [2, 3].

Based on previous studies of MgB₂ superconducting material, the addition of silicon carbide without heat treatment has been done to see superconductivity characteristics and the FWHM values to higher values, suggesting that the crystallization in the samples has been reduced [4-6]. Another attempt by Herbirowo et al. [4] has shown that the addition of nano-sized silicon carbide (SiC) into superconducting MgB₂ powder through *ex-situ* powder-in-tube process lowers the critical temperature along with the decrease in MgB₂ phase fraction. SiC has manifested in the forms of aggregates of pores and polycrystalline structures [4].

According to Erhan Aksu et al., the variation of sintered temperature has dramatically affected the formation of a single-phase MgB₂ achieved in the temperature range of 750-900°C through solid-state reactions of the elements Mg and B to produce the most optimum superconducting properties [7].

The objective of this research is to produce superconducting wire material through the *in-situ* powder-in-tube process, which refers to the success of research by Varghese et al. in increasing the full width at half maximum (FWHM) of MgB₂ and the current density [8]. In contrast, this research has the novelty of silicon carbide (SiC) doping with aim to study the effect of phase formation on carbon substitution and morphological characteristics with the motivation to improve

superconductivity properties and the low-cost Fe-sheathed SS304 tubes.

2. METHODOLOGY

2.1 Experimental procedure

The synthesis of MgB₂ superconducting wire was carried out through a solid-state reaction of powder-in-tube (PIT) process using raw materials of magnesium (98.5%) and crystalline boron (99%). The first process was to weigh magnesium and boron powder with a digital scale according to the stoichiometric calculation of Mg: B = 1: 2 and 1% SiC dopant. The powder was then mixed and crushed manually using agate mortar for 1 hour until fine evenly, then put into a stainless-steel pipe type SS304. The cross-section of SS304 was reduced using a rolling machine from 6 mm to 4 mm in diameter and cut into three parts followed by sintering treatment at 750°C, 800°C, and 850°C for 2 hour and furnace cooling [9].

Characterization using the XRD (x-ray diffraction) test instrument was used to determine with SiC doping of 1%, using a Cu source with $\lambda = 1.5406$ and a measurement area of $20:10^{\circ}-100^{\circ}$. SEM-EDS (scanning electron microscopy – energy dispersive spectroscopy) was used to analyze the MgB₂ phase transformation's surface morphology, with secondary electrons under vacuum conditions and nitrogen gas flowing with a maximum pressure of 0.5 bar. Characterization of superconductivity of MgB₂ with or without 1% SiC doping was investigated using the cryogenic magnetic instruments "Teslaron pt". This instrument used a pulse tube cryocooler to create a low temperature environment and has a 4-point-probe method to measure the samples' resistivity and 4 points probe in the sample holder. Helium gas was compressed so that the



temperature can be reduced to 1.5 K. The success of producing superconducting wire is indicated by the presence of critical temperature data at very low temperatures and the formation of the MgB₂ phase during the XRD test, of which they are projectable for Magnetic Resonance Imaging applications.

3. RESULTS AND ANALYSIS

The SS304 material's formability allowed it to be rolled for advanced superconducting wires, as shown in Figure 1.



Figure 1. Sample product a) before rolling b) after rolling

Pre-rolled tube (Figure 1.a) has a diameter of 6 mm with both inner ends tightly closed by a compressive method to prevent air entering and Mg-B spilling out when heated. The final product of 4 mm diameter MgB₂ wire (Figure 1.b) was then prepared for XRD, SEM, and cryogenic tests.

3.1 MgB₂ phase and structure formation analysis

XRD data were analyzed using Match Software based on a database from the international center for diffraction data (ICDD) PDF-4 was intended to aid automated quantitative analyses by providing key reference data. It also included several resources to complement traditional analyses, such as a complete range of data simulation programs be used to analyze neutron, electron, and synchrotron data in addition to X-ray data. The XRD results for the undoped sample are shown in Figure 2, the doped sample at 750°C in Figure 3, and the doped sample at 850°C in Figure 4.

Figure 2 shows that the Mg element has seven peaks while MgB_2 compound has one peak with a hexagonal crystal structure (Table 1). This does not mean Mg content is higher than that of MgB_2 compounds; instead XRD scanning results see Mg elements to be more dominant based on the highest peaks.



Figure 2. MgB₂ + 0wt% SiC XRD patterns for sample sintered



Figure 3. MgB₂ + 1wt% SiC XRD patterns for sample sintered at 750°C



Figure 4. MgB₂ + 1%wt SiC XRD patterns for sample sintered at 850°C

Table 1. Specimen data of MgB₂ pure sintering 750°C

Phase	COD	Space Group (Crystal Structu	reLattice Parameter (A	Density (g/cm ³)
Mg 96	5-901-3059	P63/mmc (194)	Hexagonal	a=3.2147, c=5.2203	1.72700
MgB ₂ 96	5-100-0027	Fd-3m (191)	Hexagonal	a=3.0850, c=3.5230	2.62500
Fe 96	5-901-3489	P6/mm (229)	Cubic	a=2.9330	7.35000

Table 2. Specimen data of MgB₂ + 1%wt SiC at sintering 750°C

Phase	COD	Space GroupC	rystal Structur	e Lattice Parameter (A)	Density (g/cm ³)
Mg 90	5-901-305	59 P6/mm (194)	Hexagonal	a=3.2147, c=5.2203	2.62500
MgB ₂ 90	5-100-002	27 Fd-3m (191)	Hexagonal	a=8.2966, c=3.5230	5.38600
MgO 90	5-900-681	14P6/mm (225)	Cubic	a=4.1655	5.38600
Fe 90	5-901-348	89 P6/mm (229)	Cubic	a=2.8860	5.38600
Fe4O596	5-901-424	44 P6/mm (63)	Hexagonal	a=2.8737, b=9.6940, c=12.41	165.38600

Table 3. Specimen data of $MgB_2 + 1\%$ wt SiC at sintering 850°C

Phase COD	Space Group(Crystal Structur	eLattice Parameter (A)	Density (g/cm ³)
MgB ₂ 96-100-002	27 P6/mm (191)	Hexagonal	a=3.0850, c=3.5230	2.62500
Fe3O496-900-232	22 Fd-3m (227)	Cubic	a=8.2966	5.38600

In Figure 3, Mg element has eight peaks with the presence of MgO compound in the form of cubic crystal (Table 2) due to the oxidation of the Mg with O from SS304 tube and bulk specimen, similar to previous research from Shi et al. [10]. No MgB₂ was observed.

In Figure 4, the MgB_2 compound was formed along with the impurity of cubic Fe_3O_4 (Table 3). This can be confirmed by the morphological results, which reacted directly with SS304-sheathed.

3.2 MgB₂ morphological analysis

Evolution of microstructure of pure MgB_2 wire synthesized with temperature variations of 750°C and 850°C.





Figure 5 shows the results of SEM observations of pure MgB₂ wire sintered at 750°C and 850°C. At Figure 5a, the MgB₂ phase appeared to be homogeneous. Mg and B powders were sintered, and diffusion appeared to be imperfect with the presence of a slight porosity. In Figure 5b; 6a; 6b, the amount

of porosity began to decrease, and micro aggregates coalesced, and the Mg phase was oxidized into MgO observed through the SEM mapping. The morphological image in Figure 5b shows a small grain, very compact, and a smooth interface with a metal sheath [11]. Microstructure evolution of 1% SiC-doped MgB₂ wire synthesized with the temperature variations of 750°C, 800°C, and 850°C shows at Figure 6.



Figure 6. Morphological analysis of MgB₂ + SiC wires using SEM: a) 750°C b) 800°C c) 850°C

Figure 6a shows porosity between the aggregates, which themselves look like a stack of layers. Less porosity and uniformly prolonged grain boundaries are seen in Figure 6b. However, in Figure 6c the surface of clusters looks enlarged, and the number of cavities is reduced.

3.3 Superconductivity analysis

Figure 7 shows R vs. Tc of pure MgB_2 sintered at 750°C. The sample had a Tc_{onset} of 42.91 K and Tc₀ of 33.51 K higher than previous studies [1, 12].

In Figure 8, for 1% SiC-doped MgB₂ sintered at 750°C, resistivity decreases to 0, with Tc_{onset} of 42.10 K and Tc_0 of 37.18 K. Apparently, 1% SiC addition increases Tc_0 for sintering at 750°C. This may be induced by the presence of oxides MgO and Fe₄O₅ (see Figure 3 and Table 2), leading to higher oxide layers and crystallinity, thus promoting higher Tc_0 .

In Figure 9 for 1% SiC-doped MgB_2 sintered at 800°C no superconducting properties are observed with a Tc_{onset} of 43.16 K and no Tc0. This may be correlated to the transition process of annihilation of MgO layers with higher sintering temperatures.











Figure 9. Resistance versus temperature graph of MgB₂+1wt% SiC sample sintered at 800°C



Figure 10. T vs R at 1% SiC + MgB₂ 850°C

In Figure 10 for 1% SiC-doped MgB₂ sintered at 850°C, the superconductivity occurs with Tc_{onset} 43.85 and Tc₀ 25.10 K. This Tc₀ is lower than that of pure MgB₂ (Figure 7) and 1% SiC-doped MgB₂ sintered at 750°C (Figure 8). This situation may be induced by the grain growth phenomenon thus lowering crystallinity as observed in Figure 7.

Table 4 summarizes that the most optimum superconductivity takes place in 1% SiC-doped MgB₂ sintered at 750°C. This behavior is caused by the presence of MgO (see Figure 3 and Table 2) leading to higher oxide layers and crystallinity thus promoting higher Tc₀. For pure MgB₂ sintered at 750°C and SiC-doped films MgB₂ sintered at T > 750°C, the resistivity value before the superconductor transition is very small [13].

Table 4. Critical Temperature in Each Sample

Sample	Temperature (°C)	Tc _{onset} (K)	Тс ₀ (К)	ΔT _c (K)
Pure MgB2 MgB2 +	750	42.91	33.51	9.4
SiC	750	42.10	37.18	4.92
MgB ₂ + 1% SiC	800	43.14	n/a	-
MgB ₂ + 1% SiC	850	43.85	25.10	18.75

4. CONCLUSION

This MgB₂ fabrication is a developmental stage from previous research predominated in bulk material form. In this research the innovation was made by using stainless steel casing or sheath (low cost and good corrosion resistance) yet still exhibiting superconductivity properties. 1% SiC optimally increased the critical temperature of MgB₂ along with the sintering at 750°C for one hour. This phenomenon was induced by the formation of dominant oxides of MgO along with Fe₄O₅ in MgB₂ matrix leading to higher crystallinity. Addition of 1% SiC with higher sintering temperature reduced or annihilated oxide layers along with grain growth resulting in lower Tc₀.

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CONTRIBUTORSHIP

Satrio Herbirowo: conceived and designed the research, performed the analysis, data curation, wrote the first draft of manuscript. Agung Imaduddin: supervision and conceptualization. Hendrik: discussed the results and data curation. Andika Widya Pramono: discussed the results, revised the manuscript and approved the final manuscript. Sunardi: discussed the results, read and approved the final manuscript. Iman Saefuloh: discussed the results, read and approved the final manuscript.

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