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Properties of Polyhedral Oligomeric Silsesquioxane-Modified Cellulose Insulation Paper with Different Number of Phenyls



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ABSTRACT

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Polyhedral oligomeric silsesquioxane (POSS) has a total of eight substituents. The number of substituents determines the modification effect of POSS. Through simulation of molecular dynamics, this paper explores how the number of substituents affects the POSSmodified cellulose insulation paper. Specifically, the mechanical properties, thermal stability and polarizability were calculated for cellulose models with 1-phenyl POSS, 2phenyl POSS, 3-phenyl POSS, 4-phenyl POSS, 5-phenyl POSS, 6-phenyl POSS, 7-phenyl POSS, and 8-phenyl POSS, respectively. The results show that the cellulose model modified by 6-phenyl POSS achieved better effect than the other seven models, as evidenced by its distinctively large bulk modulus, shear modulus, and elastic modulus. Besides, the cellulose model modified by 6-phenyl POSS had the most stable and the minimum mean square displacement (MSD). This is because the mechanical properties of the insulation paper are improved, as the motion of cellulose chains is inhibited by the nano-size effect of POSS and the effect of phenyls. Further, the cellulose model modified by 6-phenyl POSS also had the smallest polarizability among the modified models. Therefore, the 6-phenyl POSS-modified cellulose model can effectively reduce the polarizability of cellulose, uniform the electric field distribution in oil-paper insulation system, and thus enhance the insulation property of transformers.

1. INTRODUCTION

The development of power network has posed severe challenges to the oil-paper insulation system in power transformers. During the long-term operation, the mechanical properties and thermal stability of the transformers will be undermined by the thermal aging of the insulation paper, i.e., the decline of insulation performance [1-5]. In fact, the performance of a power transformer hinges directly on the property of insulation paper [6-10]. A popular way to improve the property of insulation paper is to modify cellulose insulation paper with nano-technology.

Nanoparticle modified insulation paper has been successfully applied in many fields [11-13]. Liao et al. [14] modified insulation paper with nano-TiO₂ of different mass fractions, immersed the modified insulation paper in the insulating oil in vacuum, and tested that nano-TiO₂ can significantly improve the strength of the power-frequency breakdown field for the insulation paper: when the mass fraction of nano-TiO₂ was 3%, the breakdown field strength increased by the greatest extent, while the dielectric constant, dielectric loss, and conductivity of the insulation paper reached the minimum. Through deposition and chemical modification, Rodríguez-Uicab et al. [15] added multiwalled carbon nanotubes to the surface of meta-aramid fiber, and verified the effectiveness of the modification method by scanning electron microscopy (SEM), Raman spectroscopy and X-ray photoelectron spectroscopy (XPS). Chen [16] modified the insulating oil with nano-TiO₂, nano-Fe₃O₄, and nano-Al₂O₃, separately, and then impregnated the insulation paper in each nano-modified insulating oil; every modified insulation paper saw improvement in the power-frequency partial discharge inception voltages and surface flashover voltages, among which the improvement by nano-TiO₂ modification was the most obvious. Tang et al. [17] modified meta-aramid insulation paper with polyphenylsilsesquioxane (PPSQ), verified improved thermal stability of the modified insulation paper through experiments and molecular simulation, and analyzed the inner mechanism of the improvement based on simulation results.

The above literature of modified cellulose insulation paper suggests that nanoparticle modification is an effective way to improve the thermal stability and mechanical properties of the insulation paper. Polyhedral oligomeric silsesquioxane (POSS) [18] is a typical organic-inorganic hybrid compound with nanoscale molecules. The POSS has a cage-shaped inorganic core composed of alternatively linked Si-O skeleton, and involves eight substitutable groups. Any change to the substituents could alter the modification results [19]. However, there is little report on how the number of substituents affects the cellulose insulation paper modified with the POSS. Peng et al. [20] prepared poly-silsesquioxane containing different functional groups by catalytic hydrolysis and condensation, tested the thermal stability of poly-silsesquioxane through thermogravimetric analysis (TGA), and demonstrated that octaphenylsilsesquioxane (OPPOSS) has the best thermal stability. Therefore, this paper decides to explore the thermodynamic properties of POSS-modified cellulose

insulation paper with different number of phenyl substituents.

2. METHODOLOGY

2.1 Model construction

This paper probes into the influence of the number of substituents on POSS-modified cellulose insulation paper through molecular dynamics simulation. Figure 1(a) shows the 1-phenyl POSS model. Previous studies have shown that fiber models have little mechanical and chemical differences in the amorphous-region, as long as the polymerization degree of fiber molecules is smaller than 10 [21]. Tang et al. [22] tested the cellulose with polymerization degree of 20, and found that the tested mechanical properties are consistent with simulation results.

To reduce model size and speed up calculation, this paper establishes a model with the fiber molecules of the polymerization degree of 10 (Figure 1(b)). The silicon atoms, oxygen atoms, hydrogen atoms, and gray atoms are silicon, oxygen, hydrogen, and carbon, respectively. After model optimization, Amorphous_Cell module was called to build a periodic amorphous model for the blending of cellulose and POSS with different number of phenyl substituents, following the method proposed by Theodorou and Suter [23]. The initial density of the model was set to 1.2g/cm³.

Wang et al. [21] held that, under the Condensed-phase Optimized Molecular Potential for Atomistic Simulation Studies (COMPASS) stand, the steady-state density of cellulose is close to the real value, reflecting the real state of cellulose. Therefore, this paper adopts the COMPASS stand, and modifies cellulose insulation paper with 1-phenyl POSS, 2-phenyl POSS, 3-phenyl POSS, 4-phenyl POSS, 5-phenyl POSS, 6-phenyl POSS, 7-phenyl POSS, and 8-phenyl POSS, separately. Among them, the 1-phenyl POSS, 4-phenyl POSS, and 8-phenyl POSS models are shown in Figure 2.



Figure 1. Sketch diagram of the established model



Figure 2. (a) Cellulose model modified with 1-phenyl POSS (b) Cellulose model modified with 4-phenyl POSS (c) Cellulose model modified with 8-phenyl POSS

2.2 Parameter setting

To keep the model in a stable state, the structure of each model was optimized by 10,000 steps, and the rationality of the model structure was estimated by adjusting the system energy and temperature and annealing them by ten 300-900K cycles with an interval of 50K in constant-temperature, constant-volume ensemble (NVT), before the calculation of molecular dynamics. If the model was found to be unreasonable, optimization and annealing would be carried out again; if the model was found to be reasonable, the model with the minimum energy would be selected, and the dynamic calculation of 500ps would be conducted under the constanttemperature, constant-pressure ensemble (NPT). To increase the model density close to the experimental value, the pressure was set to 1GPa. Then, the dynamic calculation of 500ps was performed under NVT to make the model fully balanced, and continued under NPT to ensure the model performs similarly as the real material. Under the temperature of 343K and the atmospheric pressure, the dynamic information was collected every 0.5ps to provide the trajectory information for the subsequent performance calculation. The COMPASS force field was adopted throughout the process.

3. CALCULATION

3.1 Mechanical properties

The polycrystalline substance composed of randomly oriented microcrystals is statistically isotropic. Its mechanical parameters can be obtained by Reuss average method [24] for isotropic materials. Specifically, the volume modulus (K), shear modulus (G), elastic modulus (E), Poisson's ratio (ν) and Cauchy pressures (C_{12} - C_{44}) can be respectively calculated by:

$$K = [3(a+2b)]^{-1} \tag{1}$$

$$G = 5[(4a - 4b + 3c)]^{-1}$$
(2)

$$E = 2G(1 + \nu) = 3K(1 - 2\nu)$$
(3)

$$C_{44} = \frac{1}{2}(C_{11} - C_{12}) \tag{4}$$

where,

$$a = \frac{1}{3}(S_{11} + S_{22} + S_{33}) \tag{5}$$

$$b = \frac{1}{3}(S_{12} + S_{23} + S_{31}) \tag{6}$$

$$c = \frac{1}{3}(S_{44} + S_{55} + S_{66}) \tag{7}$$

Note that C_{ij} is a 6 × 6 elastic coefficient matrix; *a*, *b*, and *c* represent the column vectors formed by the side length at the deformation time. The inverse matrix of C_{ij} needs to be computed to drive matrix S_{ij} .

In elastic mechanics, the elastic modulus (E), i.e., the ratio of stress to strain acting on the material, is positively correlated with the material's ability to resist deformation. The shear modulus (G), i.e., the ratio of shear stress to strain, is positively correlated with the material's resistance to shear stress; it is also known as rigid modulus. The volume modulus (K), i.e., the relation between the volume strain and the mean stress of the material, is positively correlated with the material's incompressibility. The Poisson's ratio (v), i.e., the ratio of transverse strain to longitudinal strain, is positively correlated with the material's plasticity; it reflects the relationship between deformation in two directions, and has little to do with the degree of deformation. The Cauchy pressures $(C_{12}-C_{44})$ reflects the ductility of the material; if the value is positive, the material has good ductility; if the value is negative, the material is highly brittle.

3.2 Mean square displacement (MSD)

The MSD, i.e., the mean particle displacement, is often used to characterize the motion performance of particles [25]. This index is positively correlated with the intensity of chain motion, and the thermal instability of cellulose. The MSD value can be calculated by:

$$MSD = \langle |r_i(t) - r_i(0)| \rangle^2 \tag{8}$$

where, $r_i(t)$ and $r_i(0)$ are the position vectors of a molecule or an atom at time t or initial time i, respectively.

3.3 Free volume

The cellulose material consists of molecules and cavities. The volume not occupied by molecules is called the free volume. The free volume fraction can be defined as the ratio of the free volume to the total volume of the material:

$$FFV = \frac{V_f}{V_f + V_0} \tag{9}$$

where, FFV is free volume fraction; V_f is free volume; V_o is occupied volume.

3.4 Dipole moments and polarization

The dielectric constant, a commonly used metric of the insulating properties of dielectric insulation materials, is mainly determined by the mean polarizability [26]. The polarizability of a polymer can be derived from the dipole moments per unit volume. The molecular dipole moments in the directions of x, y and z can be obtained by analyzing the dynamic trajectory files through molecular dynamics simulation.

4. RESULTS AND DISCUSSION

4.1 Mechanical properties

Figure 3 presents the mechanical properties of each modified cellulose model. The relevant mechanical parameters were calculated through molecular dynamics simulation.



Figure 3. Mechanical properties of each modified cellulose model

It can be seen from Figure 3 that, the 6-phenyl POSSmodified cellulose model achieved much higher bulk modulus (K), shear modulus (G), and elastic modulus (E) than the other modified cellulose models, while having similar Poisson's ratio (ν), bulk modulus, and shear modulus (K/G) as the other models. The results show that the 6-phenyl POSS-modified cellulose model has the best modification effect, as evidenced by its extraordinary anti-deformation ability and incompressibility. The main reason for the good effect is as follows:

The highly stable Si-O bond and cage structure of POSS improve the mechanical properties of the cellulose. The POSS has the advantages of stable structure and large volume. In molecular dynamics simulation, the angle of Si-O bond only changes slightly, and the three-dimensional (3D) spatial structure of POSS basically remains stable. This means the motion of POSS needs to overcome great forces. In addition, the aromatic structure of phenyl has good heat resistance and rigidity. Thus, the addition of phenyl improves the mechanical properties of the modified model. Among the eight models, the cellulose model modified by the 6-phenyl POSS witnesses the smoothest chain motion. The additive nano-particles reduces the gap between the cellulose molecular chains, inhibits the motion of these chains, and reduces the degree of motion of

the cellulose molecules, giving the modified model outstanding mechanical properties.

4.2 MSD

To study the chain motion in each modified model, the MSD of each model was calculated by the Forcite module for 200ps. The calculated results are shown in Figure 4. Note that the simulation temperature was kept at 343K to mimic the working state of cellulose insulation paper in power transformers.

As shown in Figure 4, the MSD curve of 6-phenyl POSSmodified cellulose model was smoother than any other model, while that of 5-phenyl POSS and 8-phenyl POSS modified cellulose models was steep, with a large slope. Hence, the latter two models have strong molecular chain motions. During the simulation, the MSD of the fiber chains in the 6phenyl POSS-modified cellulose model fluctuated between 0 and 1.01 Å, that of the 5-phenyl POSS-modified cellulose model fell within 0 and 1.64Å, and that in the 8-phenyl POSSmodified cellulose model ranged from 0 to 1.55Å. Overall, the 6-phenyl POSS effectively restrains the motion of cellulose chains, and leads to the best modification effect. Therefore, 6phenyl POSS has the strongest binding effect on cellulose molecular chains.



Figure 4. MSD of each modified cellulose model

4.3 Free volume

By analyzing the trajectory files, the free volume fraction of each modified cellulose model was obtained. The relevant results are reported in Table 1 and Figure 5, where the blue and gray colors represent the free volume and the occupied volume, respectively.

Table 1. Free volume fraction of each modified cellulose modified
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Model	1-phenyl POSS	2-phenyl POSS	3-phenyl POSS	4-phenyl POSS	5 phenyl POSS	6-phenyl POSS	7 phenyl POSS	8-phenyl POSS
Occupied/Å ³	16,812.59	23,573.68	21,901.16	23,652.63	26,958.92	28,801.03	32,133.99	33,715.99
Free/Å ³	2,080.16	3,077.36	3,533.42	2,776.27	3,957.15	3,669.92	4,267.97	5,178.71
FFV/%	11.01	11.55	13.89	10.50	12.80	11.30	11.72	13.31



Figure 5. Free volume of each modified cellulose model

From Figure 5 and Table 1, it can be observed that 4-phenyl POSS-modified cellulose model had the smallest free volume fraction, while of 5-phenyl POSS and 8-phenyl POSS had comparatively large free volume fractions. This echoes with the previous results on MSD. Under the nano-size effect of POSS, the remaining cellulose chains bonded more tightly, resulting in closer intermolecular connection; the free volume is also reduced as some spaces are occupied by the molecular chains.

Notably, the free volume of the 6-phenyl POSS model was

well-distributed, and relatively small within per unit volume. The cellulose molecular chains are compressed, narrowing the gap between them. As a result, the structure of the 6-phenyl POSS model is highly stable. The molecular chain motion of the model is limited, making the MSD curve more smooth.

4.4 Dipole moments and polarization

The dipole moments of 1-phenyl POSS, 2-phenyl POSS, 3-phenyl POSS, 4-phenyl POSS, 5-phenyl POSS, 6-phenyl

POSS, 7-phenyl POSS, and 8-phenyl POSS modified cellulose models in the three directions were calculated, and the results are shown in Figure 6.

The total dipole moment was calculated by the vector sum

of dipole moments in the three directions, and then the mean polarizability was obtained by dividing the total dipole moment by the total volume. The results are shown in Table 2.



Figure 6. Dipole moments of each modified cellulose model

The Model	Total volume (Å ³)	Total Dipole Moment (D)	Polarizability (D·Å ⁻³)
1-phenyl POSS	17,070.5	39.12	2.29×10 ⁻³
2-phenyl POSS	24,079.6	36.65	1.52×10 ⁻³

42.72

37.32

90.31

41.67

68.02

81.49

22,465.8

24,175.3

27,514.9

29,135.0

32,739.7

Table 2. Polarization of each modified cellulose model

8-phenyl POSS	S 34,609.0					
From Figure 6 and Table 2, the PC	SS with different number					
of phenyl substituents influenced the	e polarization of modified					
cellulose models. Comparatively,	the 6-phenyl POSS-					
modified model realized the lowest	polarizability. Because of					
the low dielectric constant of	POSS molecules, the					
polarization of the modified cellulo	se model decreased with					
the addition of POSS. In addition, the molecular weight of						
POSS is much larger than that of cellulose chain, making it						
difficult to flip the dipole bonded	by nanoparticles in the					
modified model. The higher the cont	ent of POSS, the more the					
bound dipoles, and less polarized the	model. Further, owing to					
the cage structure of POSS and the	addition of phenyls, the					
porosity and air content of the modified	fied model increased, and					
the polarization of cellulose decrea	sed, with the addition of					
POSS.						

3-phenyl POSS

4-phenyl POSS

5-phenyl POSS

6-phenyl POSS

7-phenyl POSS

8-phenvl POSS

5. CONCLUSIONS

Through molecular dynamics simulation, this paper studies the properties of the cellulose insulation paper modified by POSS with different number of phenyls, and draws the following conclusions:

1.90×10-3

1.54×10-3

3.28×10-3 1.43×10-3

2.08×10-3

2.36×10-3

(1) The number of phenyl substituents affects the mechanical properties and thermal stability of POSS-modified cellulose insulation paper.

(2) The 6-phenyl POSS-modified cellulose model shows the best mechanical properties and thermal stability, the least MSD, and the largest bulk modulus, shear modulus and elastic modulus, among all POSS-modified cellulose models.

(3) The 6-phenyl POSS-modified cellulose model also achieves the lowest polarizability among all models. This model can effectively reduce the polarization of cellulose, optimize the distribution of electric field in oil-paper insulation system, and thus enhance the insulation of transformers.

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