



Assessment of Water Quality Impacts from Tharthar and Habbaniyah Lakes Canals on the Euphrates River Using HWQI and HMPI, Western Iraq

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ABSTRACT

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In recent years, Iraqi lands have been suffering from a severe water crisis due to extreme climate changes, including high temperatures and low rainfall, as well as poor water quality resulting from high concentrations of pollutants. Therefore, there is a need to preserve and sustain water resources, the current study aimed to influence the waters of Tharthar and Habbaniyah lakes on the quality of the Euphrates River near the city of Al-Khalidiyah (Anbar Governorate) by employing a group of physical and chemical factors and heavy metals through mathematical indicators to determine the quality of water, namely the Horton's Water Quality Index (HWQI) and the heavy metal pollution index (HMPI). Four sites were selected in the study area: Site 1 was at the Euphrates River (before the mixing zone), Site 2 was selected at the Habbaniyah Canal, Site 3 was selected at the Tharthar Canal, and Site 4 was selected at the Euphrates River after the mixing zone, for the period from July 2023 to March 2024. 14 environmental factors were examined, including: pH, Alkalinity, EC, Turbidity, TDS, TSS, Hardness, Ca, Mg, DO, BOD₅, NO₃, PO₄, SO₄, most of which exceeded the Iraqi standard limits except for the following factors: pH, EC, TSS, DO, NO₃, while four heavy elements were examined: Pb, Ni, Cd, Cu, all of whose concentrations were within the Iraqi standard limits. The results of the concentrations of physical and chemical factors and heavy metals were consistent with the results of the HWQI and the HMPI, where the lowest rate of the HWQI was recorded at 74.5 in the Tharthar Canal, where the water quality was described as good, while the highest rate was recorded at 142 in the Euphrates River (site 1), where the water quality was described as poor, compared to Site 4 (Mixing Area), which scored 73.5 (Good). The highest HMPI was recorded at 6.323 in Habbaniyah Canal, while the lowest was 0.921 in the Euphrates River (site 1), compared to Site 4 (Mixing Area) which scored 2.872. The results of the current study showed that the quality of the Tharthar Canal water, which is affected by the dilution of its water from the Al-Halwa Canal, which flows into it, has a positive effect on improving the quality of the Euphrates River water. Conversely, the Habbaniyah Canal water harms the quality of the Euphrates River water in the current study area. Due to the lack of application of HWQI and HMPI water quality indices in the current study area, these indices were applied, as water quality indices are the backbone of integrated water quality management, through which abstract concepts of quality are transformed into understandable data that can be measured and analyzed, which enables decision makers to identify problems and evaluate and protect aquatic ecosystems effectively and sustainably.

1. INTRODUCTION

Water quality is one of the key indicators that determine its suitability for human use, whether for drinking, agriculture, industry, or even maintaining ecological balance [1]. With increasing environmental and human pressures on water resources due to climate change, increased consumption rates, and pollution, there is an urgent need to monitor water quality and identify indicators that reflect its condition. Water quality indices serve as assessment tools that provide accurate and measurable information about the physical, chemical, and biological properties of water, helping to determine its

compliance with established health and environmental standards. These indicators play a pivotal role in developing integrated water resource management strategies and ensuring their sustainability for future generations [2]. Lake Habbaniyah suffers from many pollutants. Its water is characterized by being hard and sulfurous, with sufficient amounts of nutrients [3]. The Warar Canal is the most influential source of water on the lake, which is the main feeder of the lake's water. The water of the Warar Canal is affected by untreated sewage from homes and the maternity and children's hospital, as well as water from agricultural and industrial and many studies have shown that its water contains

a high percentage of sulfates [2, 3].

Tharthar waters suffer from salinity and hardness due to the geological formations that lie above it, and the dead storage area is 40 meters (storage that cannot be emptied naturally, which poses a challenge in the process of using lake water as a feeder to supplement the Euphrates River water during the dry season via the Tharthar-Euphrates Canal [4]. On the other hand, the waters of this canal are affected by the Tigris River waters via the Tigris Canal and agricultural, industrial, and human uses. Heavy metals are widely present in aquatic ecosystems, especially during the current century, as a result of the expansion and excessive exploitation of natural resources by human activities, as living organisms in general and humans in particular are exposed to them. Heavy metals are elements with a high molecular weight, and their density is approximately four to five times that of water at the same pressure and temperature [5]. Some heavy elements are involved in the vital processes of living organisms, plants, and animals, such as copper and zinc, as micronutrients for them, while some metals, such as cadmium and lead, have no known physiological role. Living organisms are biologically affected in the long term when exposed to heavy metals, as these metals accumulate in their tissues, leading to damage to their functions and destruction [6]. The areas surrounding the Habbaniyah Canal suffer from the indiscriminate dumping of industrial waste, which contributes to the contamination of the area's water with lead and cadmium from auto repair shops, engines, and electric generators, as well as lead emissions from the exhaust of these engines. In addition, the high temperature of the water increases the solubility of the lead element [7]. Pesticides play a role in increasing the concentrations of lead and cadmium in the water, as is the case in the agricultural areas adjacent to the Tharthar Canal [8]. The waters of the two canals may suffer from nickel pollution from sewage and agricultural wastewater, such as phosphate fertilizers. Also, the waters of the two canals may suffer from copper pollution from agricultural waste, such as fungicides [9]. The current

study is one of the pioneering studies in the study area on the effect of the two lakes' water channels on the quality of the Euphrates River water, as it came as a continuation of the study of Al-Tamimi and Al-Lahebi [4], which dealt with the study of two types of water quality indices, namely NSF-WQI and CCME-WQI. This research aims to highlight the most important indicators of water quality selected in the current study and explain the methods of measuring them and their importance in assessing the condition of the Euphrates River in the current study sites, in addition to showing the effects of environmental changes on the Tharthar and Habbaniyah lakes in the Euphrates River, within the framework of promoting public health and achieving sustainable development, especially related to clean water and sanitation, industrial and agricultural services.

2. MATERIAL AND METHODS

2.1 Study area

Lake Tharthar is located in Iraq, northwest of Tikrit Governorate and north of Anbar Governorate. It is one of the largest natural depressions in Iraq. It has been used since 1956 to store water from the Tigris River. Tharthar was later connected to the Tigris and Euphrates Rivers, allowing sufficient quantities of irrigation water to be returned to the Tigris and Euphrates Rivers. Its area is 2,710 km², with the highest storage level reaching 65 m and the storage volume at this level reaching 85.59 billion m³. In 1976, the Tharthar-Euphrates canal was established south of Tharthar Lake and ends in the Euphrates River at the exit of Al-Dhaban (Habbaniyah Canal) to the north of the city of Fallujah. The canal is about 37.5 km long, and the Tharthar-Euphrates Canal is supplied by the Al-Helwa canal, linked to the waters of the Tigris River at the Samarra Dam and empties into the division canal, which is about 80 km long [4].

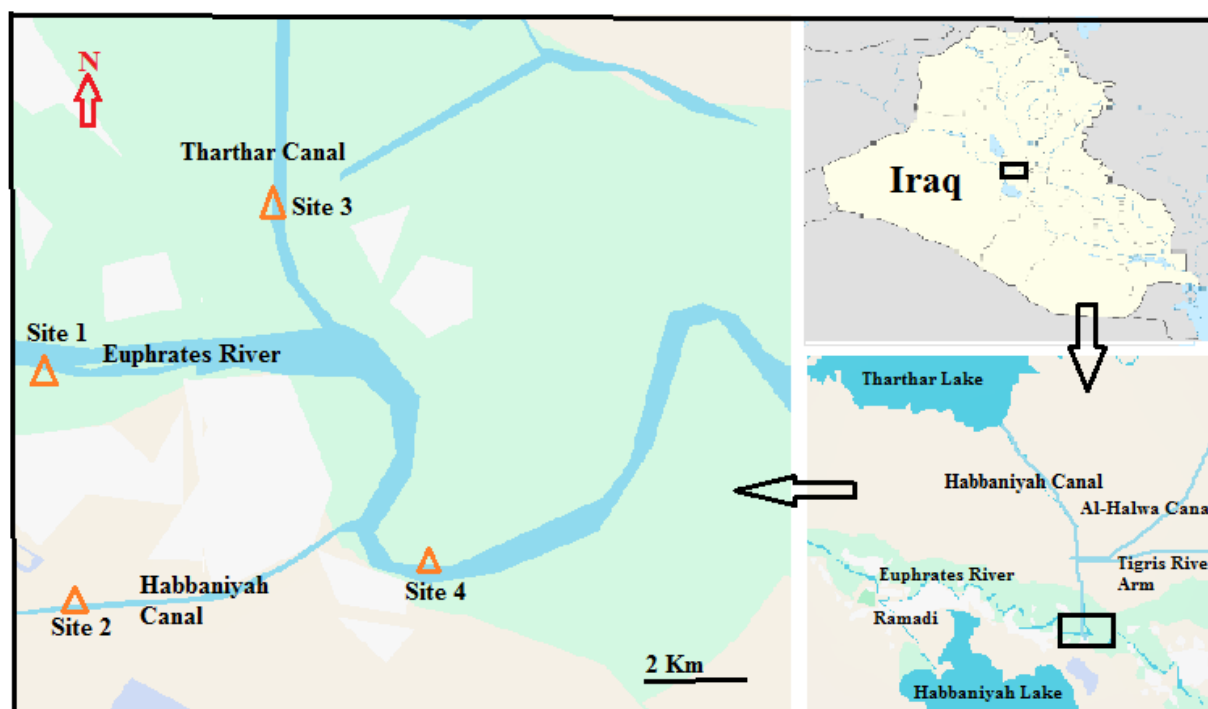


Figure 1. Map of selected sample collection sites (Google map)

Lake Habbaniyah is located in western Iraq within Anbar Governorate. It was established in 1956 to store water from the Euphrates River. It was established in 1976. The lake has a strategic water reserve. The area of the lake is about 426 Km², and the highest water storage level is 51 m. The total water capacity is about 3 billion m³. The Sin al-Dhaban Canal (Habbaniyah Canal) is an important factor in draining the lake water in the summer season at the Euphrates River near Habbaniyah District, as it returns water from Lake Habbaniyah to the Euphrates River. The length of the canal is about 9 km (Figure 1) [3].

2.2 Sampling collection

Four sites were selected during the current study period from July 2023 to March 2024 (Table 1). Site 1 was selected at the Euphrates River before the confluence of the two lakes' canal outlets in the Al-Kaldiah area. Site 2 was selected at the Sin al-Dhiban Canal to represent the waters of Lake Habbaniyah. Site 3 was selected at the Tharthar-Euphrates Canal near the al-Bushjel area to represent the waters of Lake Tharthar. Site 4 on the Euphrates River at Al-Siddiqah Bridge was chosen to represent the site of the combined effects of the waters of Lake Habbaniyah and Lake Tharthar canals (Figure 1). Water samples were collected at a rate of one sample per month for a period of nine months, and an average was taken for four seasons during the study period.

Table 1. Geographic coordinates of sample collection sites (GPS)

Site	Latitude (North)			Longitude (East)		
Site 1 (Euphrates River before mixing)	33°	23'	32"	43°	33'	56"
Site 2 (Habbaniyah Canal)	33°	22'	04"	43°	32'	42"
Site 3 (Tharthar Canal)	33°	24'	19"	43°	36'	19"
Site 4 (Euphrates River after mixing)	33°	22'	08"	43°	37'	19"

2.3 Physicochemical parameters analysis

Water samples from the studied sites were collected from a depth of 30 cm using plastic containers and were transferred directly to the laboratory to conduct the necessary tests. Field tests were conducted directly in the field, where a digital portable multi-meter (Hach HQ40d-Germany) was used to measure the pH, electrical conductivity, and total dissolved solids, while the turbidity was measured using a turbidity meter (Lovibon-Germany). Another set of factors was tested in the laboratory less than 24 hours after they were collected from the field. Where the total alkalinity (TA), total hardness (TH), calcium (CA), magnesium (Mg), total dissolved solids (TDS), total suspended solids (TSS), dissolved oxygen (DO), biochemical oxygen demand (BOD), nitrates (NO₃) and sulfates (SO₄) were measured based on the methods shown in APHA [10]. As for phosphate (PO₄), the method is shown in Eisenreich [11].

2.4 Water quality indices

2.4.1 Horton's Water Quality Index (HWOI)

According to the importance of the relevant factors, 14

factors were chosen to extract the values of this index according to the equations below [12]:

$$W_i = \frac{w_i}{\sum_{i=1}^n w_i} \quad (1)$$

where, W_i is the relative weight and w_i is the weight of each of the measured factors, so that their sum equals 1, and it was extracted according to the equation described in study [13] as follows:

$$w_i = \frac{K}{V_s} \quad (2)$$

And

$$K = 1 / \sum \frac{1}{V_s} \quad (3)$$

where, K is the constant of probability and V_s is the recommended standard value for n th parameters according to the Iraqi Standard for drinking water [14].

$$q_i = \left(\frac{V_n - V_i}{V_s - V_i} \right) * 100 \quad (4)$$

where, q_i is the parameter quality rating, V_n is the value of the measured limiting n th parameter, and V_i is the parameter ideal value, which is zero except for pH = 7 and Do = 14.6.

$$S_{li} = W_i * q_i \quad (5)$$

The S_{li} sub-index for each n th parameter.

$$WQI = \sum_{i=1}^n S_{li} \quad (6)$$

The Horton Water Quality Index scale is shown in Table 2.

Table 2. Water quality classifications according to Horton's Water Quality Index

Class	Excellent	Good	Poor	Very Poor	Unsuitable
WQI	< 50	50 - 100	100 - 200	200 - 300	> 300

2.4.2 Heavy metal pollution index (HMPI)

It is one of the modified methods of Horton's guide method, except for neglecting the value of V_i , as shown in study [5].

$$HMPI = \frac{\sum_{i=1}^n Q_i * w_i}{\sum_{i=1}^n w_i} \quad (7)$$

where, w_i is the relative weight of each of the measured i th parameter and can be deduced according to the aforementioned Eqs. (2) and (3), provided that the sum of all weights equals 1.

And Q_i is the subindex of the i th parameter. It can be extracted according to Eq. (8) as below:

$$Q_i = \frac{V_n}{S_n} * 100 \quad (8)$$

where, V_n are the values of the n th parameter during the current study, and S_n is the standard values approved by the Iraqi standard specifications for drinking water. Heavy metal pollution index values range less than 100 [15].

2.5 Statistical analysis

The statistical program SAS [16] was used to analyze the effect of sites on the studied traits, and the significant differences between the averages were compared using the least significant difference test (LSD) at the probability level of 0.05. The statistical program SPSS ver.16 was used to design the Pearson matrix to extract the correlation coefficient between all physical and chemical parameters and water quality indices at the probability level of 0.05 and 0.01.

3. RESULTS AND DISCUSSION

3.1 Physical and chemical parameters analysis

Table 3 shows the values of the physical and chemical parameters recorded at the sample collection sites during the current study period.

Table 3. Range (above) and average (below) of values of physical and chemical parameters recorded at sampling sites

Parameters	Iraqi Stand.	Site 1	Site 2	Site 3	Site 4	L.S.D
pH	6.5 - 8.5	7.4 - 8.2 7.10	7.1 - 8.6 7.57	7.1 -7.9 7.4	7.2 -8.2 7.6	0.502 NS
Alkalinity (mg CaCO ₃ .L ⁻¹)	100	83 - 1189 402.0a	89 - 1259 423.8a	70-181 125.2b	96-1089 376.5a	137.9*
EC (μ.s.cm ⁻¹)	2000	1350-1942 1808a	1842 - 1987 1900s	1319-1968 1504b	1341- 1534 1482b	217.5*
Turbidity (NTU)	5	3.2 - 7.3 5.2b	2.2-6.8 4.4b	3.6-13.0 7.8a	2.7-7.1 4.7b	2.08*
TDS (mg.L ⁻¹)	500	913 - 1238 1021.8a	9.3 - 1142 999.5a	574-1239 782.8b	895-1415 1066.3a	169.4*
TSS (mg.L ⁻¹)	1000	26 - 198 67.7	25 - 225 81.0	23.4- 210 76.8	25 - 218 87.0	37.66 NS
Total Hardness (mg CaCO ₃ .L ⁻¹)	500	489 - 580 532.5a	484-578 526.0a	235-590 423.0b	306-687 497.5ab	78.91*
Ca (mg. L ⁻¹)	50	63.1 - 137 98.13a	24.2-148 90.0a	39.4-148 62.38b	44.0-161 99.0a	22.76*
Mg (mg.L ⁻¹)	50	47 - 89.7 64.2a	43.7-83.2 51.0ab	18.3-52 39.3b	29.7-87.2 61.8a	17.52*
DO (mg.L ⁻¹)	5	4.8 - 8.3 7.23	3.1-9.2 7.25	4.7-9.8 8.0	5.1-8.8 7.3	1.69 NS
BOD ₅ (mg.L ⁻¹)	3	4.3 - 11.2 6.62	4.0-13.2 6.35	2.0-11.4 6.90	1.0-15.6 6.0	1.45 NS
NO ₃ (mg.L ⁻¹)	50	0.04 - 6.20 4.19ab	0.08-12.80 5.59a	0.06-5.20 2.60b	0.04-4.22 6.90a	2.896 *
PO ₄ (mg.L ⁻¹)	0.1	0.82 -1.81 0.70a	0.24-2.20 0.74a	0.14-0.19 0.16b	0.12-0.41 0.22b	0.287 *
SO ₄ (mg.L ⁻¹)	250	301 -562 442	340-483 430	210-498 334	329-473 392	158.09 NS

Means with different letters within the same row differ significantly from each other at the probability level * (P≤0.05) and NS (Non-significant)

Table 4. Pearson correlation values between physicochemical parameters, heavy metals, and water quality indices showed only significant correlations

	pH	EC	Tur	TH	Alk	Ca	DO	BOD	PO ₄
AIK				0.977*					
TSS	0.962*								
DO				-0.999**	-0.971*				
NO ₃								-0.995**	
PO ₄		0.985*							
SO ₄				0.989*			-0.993**		
Cd						-0.980*	-0.979*		
Co						-0.985*			
HWQI		0.990*							0.996**
HMPI			0.951*						

*. Correlation is significant at the 0.05 level; **. Correlation is significant at the 0.01 level.

The effect of the pH rates of the two lake water sites on the quality of the Euphrates River water was not observed, as there were no significant differences between all sites, as their values were between 7 and 8, and this may be attributed to the fact that Iraqi water is described as having a high buffer capacity (Figure 2) [17]. The Euphrates and Habbaniyah River sites were characterized by high alkalinity exceeding the permissible limits [14], as significant differences were recorded between all sites. The highest rate was recorded in sites 1 and 2, exceeding 400 mg.L⁻¹, while the lowest rates were recorded in site 3 (Tharthar Canal) at about 125 mg.L⁻¹. This is due to the influence of this channel by the waters of the Al-Halwa Canal, which originates from the waters of the Tigris River, leading to its dilution [18]. Seasonally, the highest alkaline rates were recorded in the fall, with the exception of Site 3. This may be attributed to rainwater, which leads to soil erosion and washing, as the Euphrates River basin is known for its calcareous soil (Figure 2) [19].

In general, the EC values did not exceed the permissible limits [14], as their values showed significant differences. The values of sites 1 and 2 (river water before mixing and Habbaniyah Canal water) were close, recording 1808 and 900 $\mu\text{s.cm}^{-1}$, respectively, while the EC values recorded lower rates in sites 3 and 4 (Tharthar Canal and Euphrates River after mixing), 1504 and 1482 $\mu\text{s.cm}^{-1}$, which gives the impression that Tharthar water has a positive effect in reducing the EC values of Euphrates River water [4]. No clear seasonal variation was recorded in EC values (Figure 3). Turbidity rates showed noticeable seasonal and site-specific variations with statistically significant differences. Tharthar Canal water (site 3) recorded the highest rates of 7.8 NTU, which exceeded the

permissible limits [14], affecting the Euphrates River waters, where site 4 recorded a rate of 4.7 NTU. This may be attributed to the fact that the Tharthar Canal suffers from the effects of sewage and agricultural water that are discharged into it without treatment, in addition to the presence of fish farms that play a role in increasing the turbidity of the water [5], as in the Kaani and Kpean Rivers in Nigeria [20]. Seasonally, the lowest rates were recorded in winter due to the dilution resulting from rain [21] (Figure 3). A negative correlation was recorded between TA and DO ($r=0.971$, $P\leq 0.05$) because the increase of TA in the water may affect the biological activity of aquatic organisms by reducing the production of DO through the process of photosynthesis (Table 4) [10].

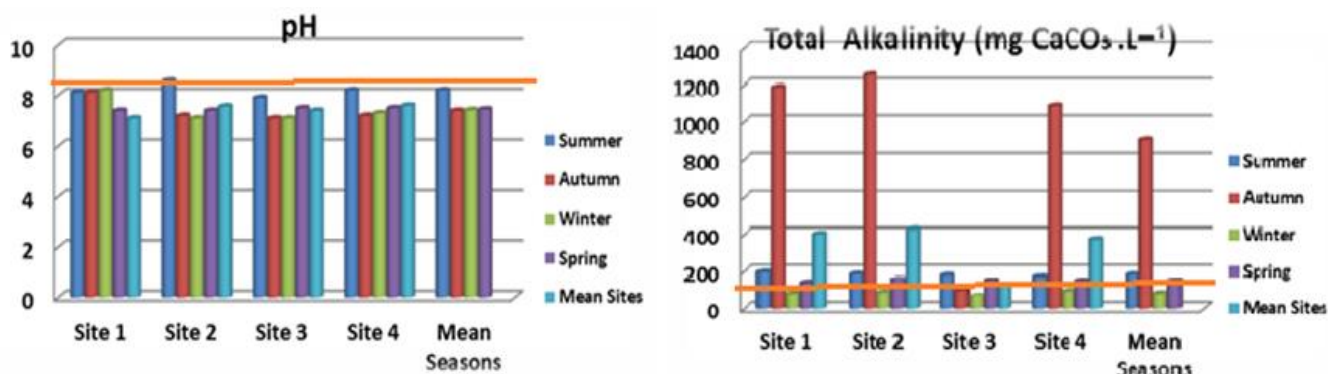


Figure 2. Seasonal and spatial variations of pH and TA values during the study period (Iraqi standard)

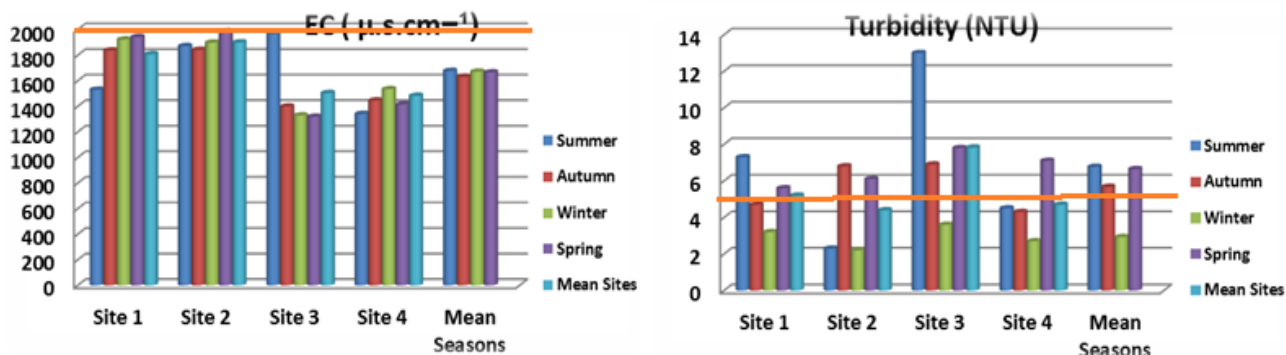


Figure 3. Seasonal and spatial variations of EC and Turbidity during the study period (Iraqi standard)

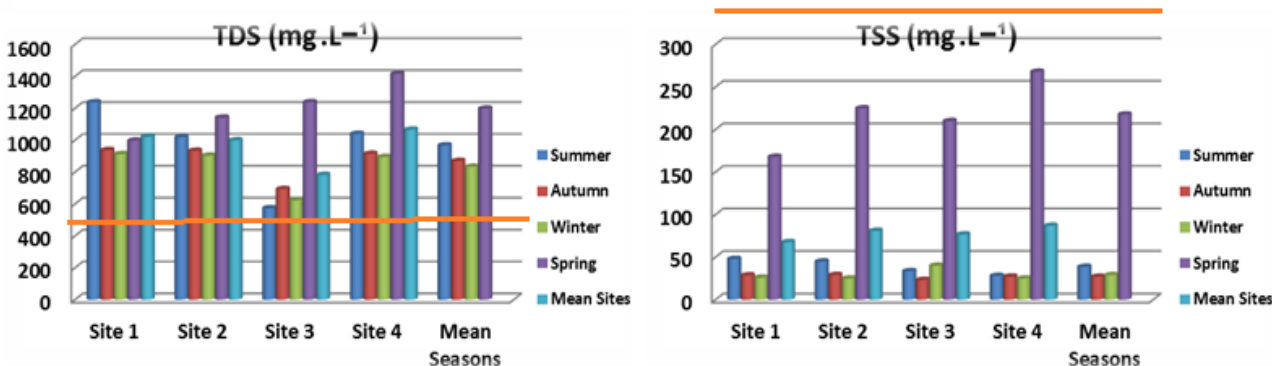


Figure 4. Seasonal and spatial variations of TDS and TSS during the study period (Iraqi standard)

TDS levels exceeded the permissible limits in all sites [14]. The lowest rates were recorded in site 3 (Tharthar Canal), 782 mg.L^{-1} due to the influence of the Al-Hilwa Canal waters and

the dilution of its water, where statistically significant differences were recorded between all sites. Seasonally, spring recorded the highest TDS values, which may be attributed to

the effect of rain and the increase in sediments washed into the rivers (Figure 4) [21]. TSS values were low and did not exceed the permissible limits at all sites, and no statistically significant differences were recorded. Seasonally, the highest values were recorded in spring, which may be attributed to an increase in suspended matter due to sediments washed into waterways (Figure 4). A statistically significant positive correlation was observed between TSS and pH ($r = 0.962$, $p \leq 0.05$) (Table 4), which may be attributed to pH-induced precipitation of dissolved constituents, consequently increasing TSS [22].

TH values were close to the approved minimum value of 500 mg.L^{-1} in all sites. The lowest mean value was 423 mg.L^{-1} in Tharthar Canal (Site 3) due to dilution by Al-Halwa Canal [18], which in turn led to a decrease in TH values of 497 mg.L^{-1} compared to sites 1 and 2 (Euphrates and Habbaniyah), which was confirmed by the statistically significant differences between all sites. Values varied seasonally, with the highest values recorded in spring due to rain and soil leaching, as the Euphrates River basin is known to be

calcareous in nature (Figure 5) [17]. Ca and Mg levels exceeded the permissible limits [14] in all sites except for site 3 (Tharthar Canal), where the lowest mean level of 39 mg.L^{-1} of Mg was recorded. In general, Ca values were higher than Mg values, which is the prevailing characteristic of Iraqi water due to the calcareous nature of the Euphrates Basin water [18]. The low Mg levels in the Tharthar Canal are due to the dilution of its water by the Al-Halwa Canal. Seasonally, Mg values rise in the summer due to drought, while the highest calcium levels are recorded in the spring due to rain, the speed of water flow, and the nature of the Euphrates Basin water (Figures 5 and 6). Record a positive correlation was recorded between TH and TA ($r=0.977$, $p \leq 0.05$) and TH with SO_4 ($r=0.969$, $p \leq 0.05$) due to their association with Ca and Mg ions [23], and a negative correlation was recorded between TH and DO ($r=-0.999$, $p \leq 0.01$) because the increase of TH in the water may affect the biological activity of aquatic organisms by reducing the production of DO through the process of photosynthesis (Table 4) [10].

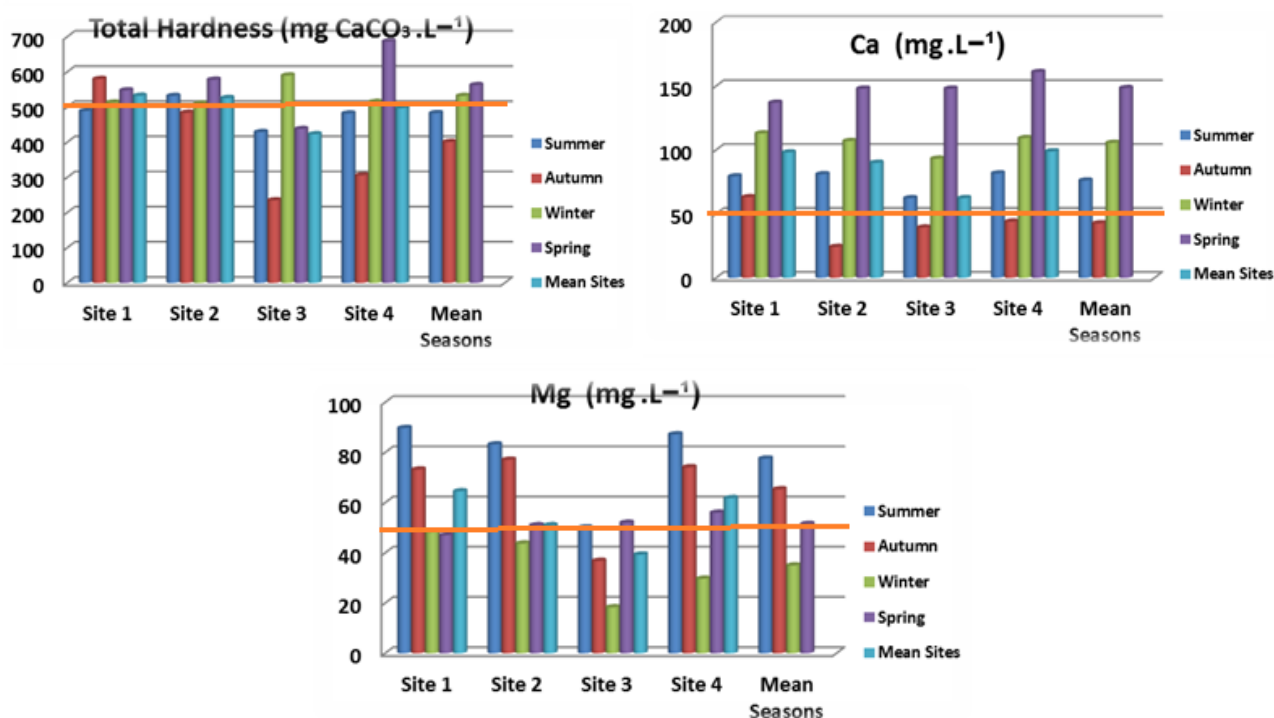


Figure 5. Seasonal and spatial variations of values TH, Ca, and Mg during the study period (Iraqi standard)

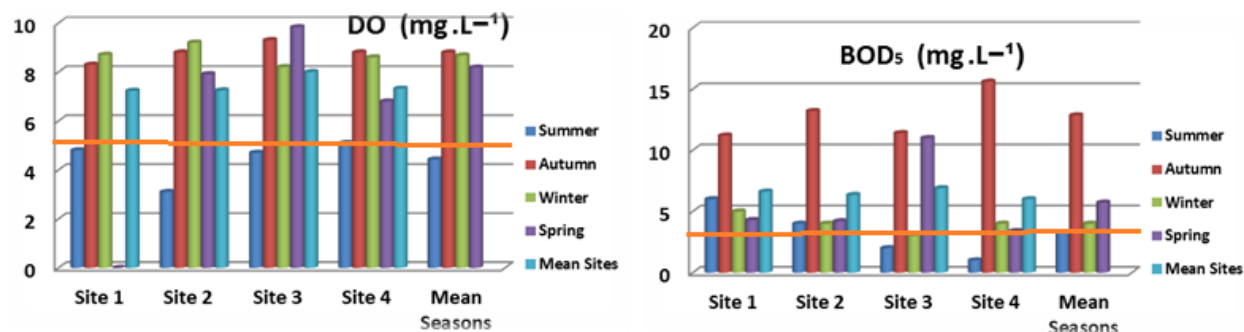


Figure 6. Seasonal and spatial variations of values DO and BOD₅ during the study period (Iraqi standard)

DO and BOD₅ values did not show significant variation among all sites, as statistically confirmed by the lack of

significant differences between sites. Site 3 (Tharthar Canal) recorded the highest mean of DO 8.0 mg.L^{-1} , while the

remaining sites recorded lower DO values. This may be attributed to the water flow velocity, with the canal being supplied with water from Al-Halwa Canal, which leads to increased surface aeration [23]. Despite the high DO levels in all sites, BOD₅ values exceeded the permissible limits [14], with the highest mean level recorded at site 3 at 6.90 mg.L⁻¹. This confirms that this increase may be due to organic pollution resulting from the exposure of all sites to multiple sources of organic pollution from sewage, industrial, and agricultural wastewater [3]. Seasonally, autumn recorded the highest DO and BOD₅ values, confirming that aeration of the Euphrates River and other sites is unable to reduce BOD₅ values due to the high organic load (Figure 6). A positive correlation was recorded between BOD₅ and NO₃ ($r=0.995$, $p\leq0.01$) (Table 4), which may be attributed to the action of biological processes. When NO₃ is available, it leads to the growth of aquatic organisms, which leads to a greater demand for DO through the respiration process [10].

NO₃ rates varied in all sites, with significant differences recorded between them. The highest rate was recorded at site

4 (Euphrates River before mixing), 6.90 mg.L⁻¹, due to the influence of agricultural land waste from the surrounding drainage water [4]. The lowest rate was recorded at site 3 (Tharthar Canal), which may be due to the dilution resulting from Al-Halwa Canal. Seasonally, the highest rate was recorded in the fall, which may be attributed to the rains, high water levels, and the speed of water flow, which leads to the rapid dissolution of DO and its availability to help in the oxidation of nitrogenous compounds to NO₃ (Figure 7) [2]. The lowest recorded rate of PO₄ was 0.16 mg.L⁻¹ at site 3 (Tharthar Channel) due to dilution by Al-Halwa Canal, while the highest rate was recorded at Habbaniyah Canal (site 2), 0.74 mg.L⁻¹ which suffers from human and industrial pollution [3], where all sites exceeded the permissible limits for PO₄ concentrations [14]. Seasonally, the highest rate was recorded in spring and autumn, which may be attributed to pollution sources (Figure 7). A statistically positive relationship was recorded between PO₄ and EC ($r=0.985$, $p\leq0.05$) (Table 4), because PO₄ may increase conductivity because it is a charged ion [24].

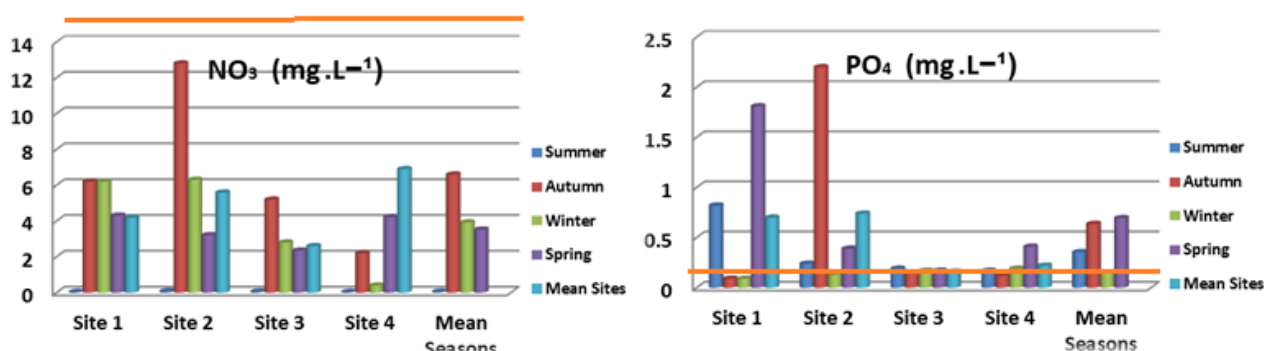


Figure 7. Seasonal and spatial variations of the values NO₃ and PO₄ during the study period (Iraqi standard)

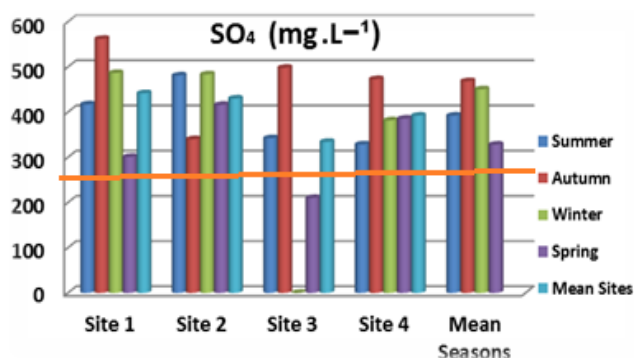


Figure 8. Seasonal and spatial variations of values SO₄ during the study period (Iraqi standard)

The Euphrates River Basin waters are characterized by high concentrations of SO₄ [19]. This case was also recorded in the Tigris River (Iraq) [14], as their values exceeded the permissible limits in all locations [14]. The lowest rates were recorded in site 3 (Tharthar Channel) at 334 mg.L⁻¹ due to its influence by Al-Halwa Channel, while the other locations recorded slightly higher than site 3. There was no statistically significant difference recorded between all locations (Figure 8). A positive correlation between SO₄ and DO ($r=0.993$, $p\leq0.01$) (Table 4) may be attributed to the biological role of SO₄, which in turn affects the production of DO by aquatic plants [23].

3.2 Water quality indices

3.2.1 Horton's Water Quality Index (HWQI)

In this study, the Horton Water Quality Index was applied to determine the extent of the influence of the Habbaniyah and Tharthar Lakes channels on the water quality of the Euphrates River in the current study area. Tables 5, 6, 7, and 8 indicate the method of extracting the values of this index in all locations. Table 8 indicates the seasonal water quality and rates of each site.

The results of HWQI showed a similarity in the rates of Site 1 (Euphrates River before mixing) and Site 2 (Habbaniyah Canal), as Site 1 recorded values ranging between 221 – 63.5 (Table 5) and an average of 142 (poor) (Table 9), as this site was considered excellent in comparison with the rest of the sites. As for site 2, it recorded values close to site 1, as its values between 77.99 (Dry) to 214.3 (Wet) (Table 6) and averaged 146 (poor) (Table 9), as this site has a direct impact of pollution from human, agricultural, and industrial activities, which was reflected in the poor quality of its water [3]. Given that the HWQI results depend on values exceeding the standard limits [14], most of the factors were recorded to exceed the standard limits for both sites, namely TDS, Mg, Ca, BOD₅, SO₄, and PO₄ (Tables 5 and 6), which negatively affected the poor quality of these sites and their high HWQI scores.

HWQI values at site 3 showed the lowest values compared to all sites, ranging between 78.2-70.6 (Table 7) with an average of 74.5 (Good) (Table 9). This result was consistent

with recording low values for most of the physical and chemical factors at this site, with a smaller number of factors exceeding the permissible limits, namely TDS, Ca, BOD₅, and SO₄ (Table 7), which was positively reflected in the decrease in the index values and the improvement in the quality of the canal water. This improvement came as a result of the dilution of the Tharthar Canal water with the water of the Al-Halwa Canal [4].

The HWQI values at site 4 (Euphrates River after mixing)

were close to those at site 3 (Tharthar Canal), ranging between 69.2 -78.1 (Table 8) with an average of 73.5 (good) (Table 9). When comparing the values of this site with those of site 1 (Euphrates River before mixing), which recorded poor water quality. The water quality of Site 4 (mixing area) is dominated by the good water quality of the Thartharo Canal inlet, which offsets the negative impact of the water quality of the Habbaniya Canal inlet and a portion of the Euphrates River water (Site 1) upstream.

Table 5. How to calculate and extract values of HWQI for site 1 (Euphrates River - before mixing)

Parameters	Values		Standard	1/s	W	Vn / vn		W*Q	
	Dry	Wet				Dry	Wet	Dry	Wet
pH	7.75	8.15	9	0.11	0.038	9.37	14.3	1.905	0.095
EC	1737	1880	2000	0.0005	0.0001	86.85	94	0.0165	0.016
Turbidity	6.45	3.95	5	0.2	0.068	129	79	5.761	6.1719
TH	518	546	500	0.002	0.00007	103.6	109.2	0.0761	0.0681
TA	168	636	500	0.002	0.00007	33.6	127.2	0.0237	0.0924
TDS	1118	925	500	0.002	0.0007	223.6	185	0.1481	0.1259
TSS	108	27	1000	0.001	0.0003	10.8	2.7	0.0046	0.00009
Mg	68	60	50	0.02	0.007	136	120	1.0425	0.823
Ca	108	88	50	0.02	0.007	216	176	1.563	0.905
DO	6	8.5	5	0.2	0.007	120	170	7.572	12.34
BOD	5.15	8.1	3	0.333	0.114	171.66	270	15.622	32.77
NO ₃	2.17	6.2	50	0.02	0.0068	4.34	12.4	0.0226	0.131
PO ₄	1.32	0.09	0.5	2	0.685	264	18	43.89	160.49
SO ₄	359	524	250	0.004	0.0013	143.6	209.6	0.245	0.2255
Sum				2.915944	1.000	1652.43	1587.47	214.2695	77.90232
				K=0.3429				HWQI= $\Sigma W*Q$	

Table 6. How to calculate and extract values of HWQI for site 2 (Al-Dhaban Canal - Habbaniyah Lake)

Parameters	Values		Standard	1/s	W	Vn / vn		W*Q	
	Dry	Wet				Dry	Wet	Dry	Wet
pH	8	7.05	9	0.11	0.038	50	2.5	1.905	0.095
EC	1929	1870	2000	0.0005	0.0001	96.45	93.5	0.0165	0.016
Turbidity	4.2	4.5	5	0.2	0.068	84	90	5.761	6.1719
TH	555	497	500	0.002	0.00007	111	99.4	0.0761	0.0681
TA	173	674	500	0.002	0.00007	34.6	134.8	0.0237	0.0924
TDS	1080	918	500	0.002	0.0007	216	183.6	0.1481	0.1259
TSS	135	27	1000	0.001	0.0003	13.5	2.7	0.0046	0.00009
Mg	76	60	50	0.02	0.007	152	120	1.0425	0.823
Ca	114	66	50	0.02	0.007	228	132	1.563	0.905
DO	5.52	9	5	0.2	0.007	110.4	180	7.572	12.34
BOD	4.1	8.6	3	0.333	0.114	136.6	286.66	15.622	32.77
NO ₃	1.65	9.55	50	0.02	0.0068	3.3	19.1	0.0226	0.131
PO ₄	0.32	1.17	0.5	2	0.685	64	234	43.89	160.49
SO ₄	448	411	250	0.004	0.0013	179.2	164.4	0.245	0.2255
Sum				2.915944	1.000	1479.117	1742.667	77.90232	214.2695
				K=0.3429				HWQI= $\Sigma W*Q$	

Table 7. How to calculate and extract values of HWQI for site 3 (Tharthar Lake Canal)

Parameters	Values		Standard	1/s	W	Vn / vn		W*Q	
	Dry	Wet				Dry	Wet	Dry	Wet
pH	8	7.1	9	0.11	0.038	50	5	1.902	0.1905
EC	1643	1363	2000	0.0005	0.0001	82.15	68.15	0.014	0.011
Turbidity	10.4	5.3	5	0.2	0.068	208	106	14.266	7.27
TH	433	412	500	0.002	0.00007	86.6	82.4	0.0593	0.0565
TA	166	84	500	0.002	0.00007	33.2	16.8	0.0227	0.01152
TDS	906	659	500	0.002	0.0007	181.2	131.8	0.124	0.0904
TSS	117	32	1000	0.001	0.0003	11.7	3.2	0.00401	0.001
Mg	51.05	28	50	0.02	0.007	102.1	56	0.7002	0.384
Ca	105	66	50	0.02	0.007	210	132	1.443	0.905
DO	7.3	8.8	5	0.2	0.007	146	176	10.01	12.07
BOD	6.5	7.2	3	0.333	0.114	216.66	240	24.768	27.43
NO ₃	1.21	4	50	0.02	0.0068	2.42	8	0.0165	0.0548
PO ₄	0.18	0.16	0.5	2	0.685	36	32	24.69	21.94
SO ₄	276	392	250	0.004	0.0013	110.4	156.8	0.1514	0.215
Sum				2.915944	1.000	1476.437	1214.15	78.17863	70.64676
				K=0.3429				HWQI= $\Sigma W*Q$	

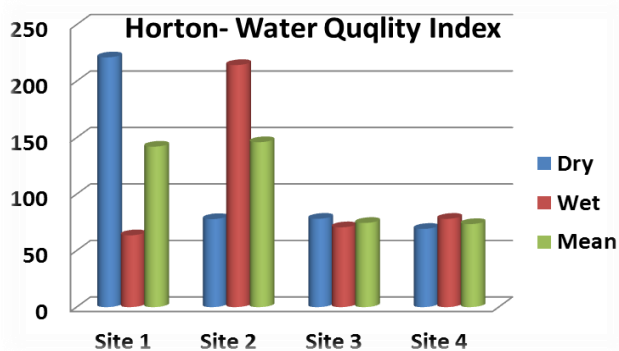
Table 8. How to calculate and extract values of HWQI for site 4 (Euphrates River - after mixing)

Parameters	Values		Standard	1/s	W	Vn / vn		W*Q	
	Dry	Wet				Dry	Wet	Dry	Wet
pH	7.85	7.25	9	0.11	0.038	42.5	12.5	1.619	0.476
EC	1378	1490	2000	0.0005	0.0001	68.9	74.5	0.0118	0.01277
Turbidity	5.8	3.45	5	0.2	0.068	116	69	7.956	4.732
TH	584	450	500	0.002	0.00007	116.8	90	0.0801	0.0617
TA	161	592	500	0.002	0.00007	32.2	118.4	0.022	0.0812
TDS	1228	905	500	0.002	0.0007	145.6	181	0.168	0.124
TSS	27.5	26.1	1000	0.001	0.0003	2.75	2.61	0.00009	0.00008
Mg	71.6	51.9	50	0.02	0.007	143.6	103.8	0.982	0.7119
Ca	121	77	50	0.02	0.007	242	154	1.6598	1.056
DO	5.95	8.7	5	0.2	0.007	119	174	8.162	11.934
BOD	2.2	10	3	0.333	0.114	73.333	333.33	8.383	38.104
NO ₃	2.13	1.3	50	0.02	0.0068	4.26	2.6	0.0292	0.0178
PO ₄	0.29	0.15	0.5	2	0.685	58	30	39.781	20.57
SO ₄	306	427	250	0.004	0.0013	122.4	170.8	0.1679	0.234
Sum				2.915944	1.000	1386.943	1516.543	69.02459	78.12558
				K=0.3429				HWQI= $\sum W*Q$	

Table 9. Horton Water Quality Index values and rates for modeling sites, and determine their water quality

Sites	Dry	Categories	Wet	Categories	Mean	Categories
1 (Euphrates River – Before mixing)	221.04	Very poor	63.546	Good	142	Poor
2 (Habbaniyah Canal)	77.902	Good	214.270	Very poor	146	Poor
3 (Tharthar canal)	78.177	Good	70.647	Good	74.5	Good
4 (Euphrates River – After mixing)	69.246	Good	78.126	Good	73.5	Good

Seasonally, the sites varied in HWQI values and did not follow a consistent pattern between the dry and wet seasons, which confirms that there is an external factor that plays a role in causing this variation, which is the pollution factor that occurs in water sources (Figure 9) [6]. A statistically positive relationship was recorded between EC, PO₄, and HWQI (Table 4), because the increase in PO₄ and EC reduces the efficiency of water quality [22]. A statistically significant relationship was recorded between the HWQI and both EC and PO₄ ($r=0.990$, $p\leq 0.05$) (Table 4) as their concentrations were consistent with the HWQI values in all sites [23].

**Figure 9.** Horton's Water Quality Index values at the sampling sites

3.2.2 Heavy metal pollution index (HMPI)

In the current study, the four heavy metals Pb, Ni, Cd, and Co were studied in the waters of the Euphrates River and the canals of the two lakes affected by them. Table 10 indicates the recorded values of these elements in all sites, with their values varying between locations by recording significant differences between all sites for each element. The table shows that all the studied heavy elements did not exceed the Iraqi standard limits [14].

Lead (Pb) is a very toxic metal if it exceeds the natural limits in aquatic systems, as it accumulates in the bodies of aquatic organisms and thus affects the primary production process of the food chain, such as algae, in addition to its harmful effects on human health. The highest rate, 0.55 $\mu\text{g.L}^{-1}$ was recorded in site 1, and the lowest rate, 0.11 $\mu\text{g.L}^{-1}$ in sites 2 and 3. The reason for the increase is due to waste from military operations near the site and from industrial sites for car repair, and from burning car fuel containing ethyl lead, which led to its increase. As for the decrease, it is due to the process of its consumption by aquatic organisms, especially algae, which carry out a process of bioremediation in the water to a large extent [24, 25] (Table 10).

Nickel (Ni) is found in two forms: first, naturally, as igneous and sedimentary rocks, and second, as a result of industrial waste that is discharged directly into the river. The highest rate, 0.26 $\mu\text{g.L}^{-1}$ was recorded in site 2 (Habbaniya Canal), which suffers from pollution due to human activities, especially the sight of car and heavy equipment repair shop and their waste, in addition to the abundance of agricultural fertilizers containing nickel. The lowest rate, 0.012 $\mu\text{g.L}^{-1}$ was in site 4, which may be attributed to the dilution of water caused by site 3 (Tharthar Canal) (Table 10) [26].

Cadmium (Cd) contamination of water poses a greater health risk because it accumulates in the tissues of all living organisms and enters the water through paint factory waste and car exhaust [27]. The lowest Cd level was recorded at 0.137 $\mu\text{g.L}^{-1}$ in site 4, and the highest level was 0.27 $\mu\text{g.L}^{-1}$ in site 2 (Al-Habbaniyah Canal) due to the presence of many car repair factories and its proximity to agricultural areas (Table 10). A positive correlation was recorded between Cd with Ca ($r=0.980$, $p\leq 0.05$) and with DO ($r=-0.979$, $p\leq 0.05$) (Table 4) due to the chemical similarity between them, as well as the relationship between their joint absorption by aquatic organisms and their joint accumulation within their tissues. Also, increasing the concentration of DO can increase the activity of aquatic organisms, which in turn leads to the

availability of Cd [28]. Cobalt (Co), this element is characterized by its very strong hardness and is found in many ores, but you cannot obtain it on the surface of the Earth because the high levels of chlorine and oxygen in the atmosphere prevent its formation. The highest rate of 0.48

$\mu\text{g.L}^{-1}$ was recorded in site 3, and the lowest rate of $0.23 \mu\text{g.L}^{-1}$ in site 1 (Table 10). A positive correlation between Ca and Co ($r=-0.985$, $p \leq 0.05$) (Table 4) may be attributed to the vital role of Ca in the aquatic environment that affects the availability of Co [23].

Table 10. Range (above) and average (below) of values of HMPI and LSD recorded at sampling sites

Heavy Metal	Site 1	Site 2	Site 3	Site 4	LSD
Pb ($\mu\text{g.L}^{-1}$)	0.0-0.09 0.55 a	0.06-0.22 0.11 b	0.0-0.26 0.11 b	0.01-0.56 0.29 ab	0.398 *
Ni ($\mu\text{g.L}^{-1}$)	0.0-0.81 0.21 a	0.10-0.52 0.26 a	0.05-0.70 0.21 ab	0.0-0.03 0.012 b	0.167 *
Cd ($\mu\text{g.L}^{-1}$)	0.0-0.09 0.032	0.01-0.75 0.27	0.0-0.15 b	0.01-0.31 0.137 ab	0.148 *
Co ($\mu\text{g.L}^{-1}$)	0.05-0.24 0.23 b	0.31-0.61 0.40 ab	0.25-0.81 0.48 a	0.0-0.094 0.24 b	0.229 *

Table 11. Evidence of HMPI values in the waters of Site 1 (Euphrates River - before the mixing process)

Heavy Metal	Mean Con. (V_i)		Iraqi Standard (S_i)	W_i	Sub Index (Q_i)		$Q_i \times W_i$		Mean HMPI $(\Sigma Q_i \times W_i)$	
	Dry	Wet			Dry	Wet	Dry	Wet	Dry	Wet
	Pb($\mu\text{g/L}$)	0.045	0.065	10	0.216	0.45	0.65	0.097	0.140	1.194
Ni($\mu\text{g/L}$)	0.085	0.13	100	0.022	0.09	0.13	0.002	0.003		
Cd ($\mu\text{g/L}$)	0.045	0.02	3	0.720	1.50	0.66	1.079	0.479		
Co ($\mu\text{g/L}$)	0.186	0.29	50	0.043	0.37	0.58	0.016	0.025		
Sum	1.00								Mean= 0.921	

Table 12. Evidence of HMPI in the waters of Site 2 (Al-Dhaban Canal-Habbaniyah Lake)

Heavy Metal	Mean Con. (V_i)		Iraqi Standard (S_i)	W_i	Sub Index (Q_i)		$Q_i \times W_i$		Mean HMPI ($\Sigma Q_i \times W_i$)	
	Dry	Wet			Dry	Wet	Dry	Wet	Dry	Wet
	Pb(μg/L)	0.14	0.09	10	0.216	1.40	0.90	0.302	0.194	4.295
Ni(μg/L)	0.51	0.21	100	0.022	0.51	0.21	0.011	0.005		
Cd (μg/L)	0.165	0.38	3	0.720	5.50	12.66	3.956	9.112		
Co (μg/L)	0.285	0.460	50	0.043	0.57	0.92	0.025	0.040		
Sum	1.00								Mean= 6.823	

Table 13. Evidence of HMPI in the waters of Site 3 (Tharthar Lake Canal)

Heavy Metal	Mean Con. (V_i)		Iraqi Standard (S_i)	W_i	Sub Index (Q_i)		$Q_i \times W_i$		Mean HMPI $(\Sigma Q_i \times W_i)$	
	Dry	Wet			Dry	Wet	Dry	Wet	Dry	Wet
	Pb($\mu\text{g/L}$)	0.09	0.13	10	0.216	0.90	1.30	0.194	0.281	2.159
Ni($\mu\text{g/L}$)	0.05	0.41	100	0.022	0.05	0.41	0.001	0.009		
Cd ($\mu\text{g/L}$)	0.08	0.06	3	0.720	2.67	2.00	1.918	1.439		
Co ($\mu\text{g/L}$)	0.53	0.39	50	0.043	1.06	0.78	0.046	0.034		
Sum	1.00								Mean = 1.961	

Table 14. Evidence of HMPI values in the waters of Site 4 (Euphrates River-Mixing Site)

Heavy Metal	Mean Con. (V_i)		Iraqi Standard (S_i)	W_i	Sub Index (Q_i)		$Q_i \times W_i$		Mean HMPI $(\Sigma Q_i \times W_i)$	
	Dry	Wet			Dry	Wet	Dry	Wet	Dry	Wet
	Pb (µg/L)	0.11	0.475	10	0.216	1.100	4.750	0.237	1.025	2.637
Ni (µg/L)	0.015	0.01	100	0.022	0.015	0.010	0.0003	0.0002		
Cd (µg/L)	0.10	0.085	3	0.720	3.330	2.833	2.398	2.038		
Co (µg/L)	0.01	0.492	50	0.043	0.020	0.984	0.0008	0.0425		
Sum				1.00					Mean= 2.872	

Tables 11, 12, 13, and 14 show the values of the HMPI index and the method of extracting them mathematically. Given that the concentrations of the four heavy elements measured in the current study did not exceed the permissible standard limits, low values were recorded for the HMPI [29].

The highest rate of HMPI was recorded at site 2 (Habbaniyah Canal), about $6.823 \mu\text{g.L}^{-1}$ (Figure 10) due to industrial and agricultural pollution and war waste that falls along the path of this canal [5, 13]. Seasonal variations in HMPI values did not show a clear pattern across all sites,

which may be attributed to the low concentrations recorded that prevented any variation from appearing. Based on the current results, the studied sites of the Euphrates River and the Tharthar and Habbaniyah lakes canals are considered safe sites with regard to their exposure to heavy metals. It was statistically recorded that there is a positive relationship between the HWPI with turbidity ($r=0.951$, $p\leq 0.05$) (Table 4). This may be due to the heavy elements being attached to the suspended particles in the water, which increases their concentration with an increase in the turbidity of the water [27].

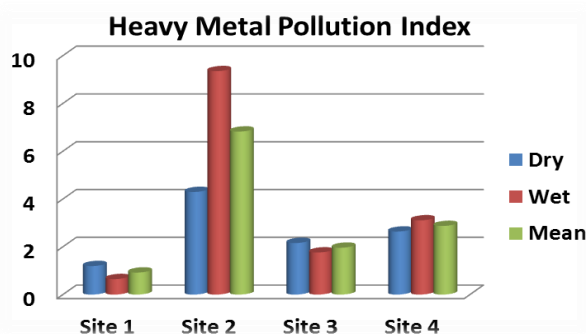


Figure 10. HMPI values at the sampling sites

4. CONCLUSION

The HWQI and the HMPI demonstrated their effectiveness in such environmental studies, as their results were consistent with the results of physical and chemical tests and heavy metal concentrations. The Tharthar Canal plays an important role in improving and mitigating the water quality of the Euphrates River in the current study area, as recorded concentrations of most parameters were lower than those of river water, including TA, EC, TDS, Mg, NO_3 , PO_4 , and SO_4 . Although the water of Lake Tharthar is characterized by high salt concentrations and high hardness due to its geological nature and the fact that it is a closed body of water, the Al-Halwa Canal plays a significant role in improving the water quality of the Tharthar Canal. The water of the Euphrates River and the waters of the Tharthar and Habbaniyah canals have safe concentrations of heavy metals measured in the current study. We recommend the necessity of activating the Environmental Protection Law and periodic monitoring to detect and address pollution sources at the current study sites.

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