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Green Synthesis of Zinc Oxide Nanoparticles Using Conocarpus lancifolius Extract and Their Antibacterial Activity Against Xanthomonas citri



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ABSTRACT

The increasing resistance of pathogenic bacteria to conventional antibiotics poses a serious threat to both human health and agriculture, prompting the search for alternative antimicrobial strategies. This study aimed to investigate the antibacterial potential of zinc oxide nanoparticles (ZnO-NPs) synthesized through a green method. Xanthomonas citri, a bacterial pathogen, was isolated from infected plant leaves collected from Baghdad, Babylon, and Kut in Iraq. Zinc oxide nanoparticles were biologically synthesized using Conocarpus lancifolius leaf extract. The biologically synthesized nanoparticles were characterized using Fourier-transform infrared spectroscopy (FTIR), ultraviolet-visible spectroscopy (UV-Vis), and atomic force microscopy (AFM). The antimicrobial activity of the biologically synthesized nanoparticles against Xanthomonas citri was evaluated using the streaking technique. The results of the characterization of the ZnO-NPs revealed an absorption peak at 292 nm and an average particle size of 73.55 nm. Furthermore, the ZnO-NPs demonstrated an 18 mm clear zone against the test bacterial pathogen. At 100 mg/mL, however, the crude C. lancifolius extract had no inhibitory effect, suggesting that the formulation of the nanoparticles greatly increased the antimicrobial activity. These findings highlight the promising role of biologically synthesized ZnO-NPs as an effective alternative approach to combat bacterial resistance, with potential applications in both medical and agricultural fields.

1. INTRODUCTION

Citrus fruits are among the most extensively grown and commercially important fruit varieties in the world. They belong to the Rutaceae family [1]. Citrus species, which originated in Southeast Asia, have effectively spread to other areas because of their adaptability to a variety of soil types and climates as well as their numerous uses in the food industry, medicine, and human nutrition. Beyond their appealing flavor, citrus fruits are recognized for their richness in essential nutrients and biologically active compounds that play a crucial role in promoting health and preventing diseases [2]. Despite their high nutritional value and commercial importance, citrus plants are prone to several diseases caused by pathogens, including bacteria, fungi, viruses, and phytoplasmas. These diseases can lead to considerable reductions in fruit yield and quality, resulting in substantial economic losses for the global citrus industry [3].

One of the most serious diseases affecting citrus is citrus canker, which is triggered by the bacterium *Xanthomonas citri* subsp. *citri*. This disease causes premature fruit drop and leaf loss, which significantly reduces crop yield. It appears as elevated lesions on plant surfaces, usually surrounded by a yellow halo [4]. Highly contagious, citrus canker spreads rapidly through wind-blown rain and contaminated tools. Citrus production has suffered greatly as a result of the disease

in nations like Brazil and the United States. Strategies to manage and limit its spread include the application of copperbased antibacterial agents, adherence to strict sanitation practices, and, in some regions, the removal of infected trees as a containment measure [5]. Many citrus species are highly vulnerable to citrus canker, a severe plant disease caused by the rod-shaped, Gram-negative bacterium Xanthomonas citri. This microorganism, formally identified as Xanthomonas citri subsp. citri belongs to the family Xanthomonadaceae and is notable for its ability to spread rapidly and thrive in warm, humid climates [6]. Infection symptoms are evident on both upper and lower surfaces of citrus leaves, typically presenting as raised, blister-like lesions encircled by a distinct yellow halo. In advanced cases, significant leaf drop can occur, impairing the plant's photosynthetic efficiency [7]. On stems and branches, similar lesions may develop, often resulting in cracks and eventual dieback of the affected limbs [8]. According to studies, nanoparticles have been shown to control diseases brought on by Xanthomonas species. Biosynthesized zinc oxide nanoparticles from mangosteen peel ethanol extract effectively inhibited Xanthomonas oryzae and enhanced rice growth, highlighting their potential as ecofriendly agricultural nanomaterials [9]. Plant-based ZnO nanoparticles effectively control bacterial leaf blight, showing strong antibacterial activity, mechanistic insights, and good biocompatibility, offering a sustainable nanobiocontrol approach for crops [10].

Plants from the Conocarpus genus, which belongs to the Combretaceae family, are widely recognized for their ability to endure harsh environmental conditions, making them valuable in urban landscaping and ecological rehabilitation projects. Notably, species such as Conocarpus erectus and Conocarpus lancifolius are native to coastal regions in tropical and subtropical zones and exhibit strong tolerance to salinity. drought, and extreme heat [11]. Scientific investigations have revealed that various parts of Conocarpus, including leaves and bark, possess antibacterial, antifungal, and antioxidant properties, suggesting their potential utility in medicinal applications alongside environmental use [12]. In recent years, the eco-friendly synthesis of zinc oxide nanoparticles (ZnO-NPs) using plant materials has garnered significant attention, especially for biomedical purposes [13]. Green ZnONP synthesis is preferred because it is easy, affordable, and safe, allowing for a variety of uses [14]. These nanoparticles have demonstrated promise in various biomedical roles, including drug delivery, anticancer, antidiabetic, and agricultural uses [15]. Zinc oxide nanoparticles exhibit significant promise as antimicrobial agents in light of these advantages.

The present research aimed to synthesize ZnO-NPs using an eco-friendly green synthesis approach with *Conocarpus lancifolius* leaf extract as the reducing agent and to evaluate their antibacterial activity against *Xanthomonas citri*, the causal agent of citrus canker.

2. MATERIALS AND METHODS

2.1 Isolation of bacterial strains

Citrus leaves exhibiting symptoms resembling bacterial canker were collected from three different Iraqi provinces: Baghdad, Babylon, and Kut. These leaf samples were selected at random from different orchard trees and rows. After collection, the leaves were placed in plastic bags and transported to the laboratory for further analysis. To get rid of surface impurities, the samples were first rinsed for about ten minutes under running tap water. One lesion and roughly 12 mm of nearby bark were removed and finely chopped with a sterile scalpel. The resulting tissue fragments were macerated in phosphate-buffered saline (PBS, pH 7.2) and left to rest at ambient temperature for 10 to 20 minutes [16]. To culture the target bacteria, the processed samples were inoculated onto nutrient agar and incubated at 37°C for 48 hours. Subsequently, colonies from these cultures were streaked onto general-purpose media and incubated again for an additional 48 hours at the same temperature. The identification of the bacterial isolates was based on several diagnostic criteria, including microscopic analysis, biochemical tests, and morphological assessment of the colonies, evaluating features such as shape, size, margin, consistency, and color.

2.2 Plant collection and pretreatment

In September 2024, *C. lancifolius* leaves were collected from local markets in Baghdad, ensuring that environmental factors were considered, such as avoiding plants treated with pesticides. The plant was botanically identified as *Conocarpus lancifolius* at the Department of Life Sciences, College of Science for Girls, University of Baghdad, based on its morphological characteristics.

2.3 Preparation of the aqueous extract

Leaves from *C. lancifolius* were carefully detached from the plant, thoroughly rinsed with water, and then chopped into small fragments [17]. To improve the homogeneity of the mixture, 25 grams of the prepared leaves were combined with 100 milliliters of distilled water and sonicated for 15 minutes in an ultrasonic bath. The ultrasonic treatment aided in better extraction through improved dispersion. Afterward, the mixture was filtered using Whatman filter paper to remove solid residues. The resulting aqueous extract was divided into 10 mL portions, which were placed in centrifuge tubes and centrifuged at 400 rpm for 10 minutes to separate the supernatant from the sediment. Following the nanoparticle synthesis process, the filtrate was preserved for further use. Zinc oxide nanoparticles were subsequently biosynthesized using this extract [18].

2.4 Green synthesis of zinc oxide nanoparticles

Zinc oxide nanoparticles were synthesized through an ecofriendly method by mixing 0.5 grams of zinc chloride (ZnCl₂) powder with 10 mL of the aqueous extract derived from Conocarpus lancifolius. The ratio between ZnCl₂ and the plant extract was determined based on the protocol outlined in a recent study [19]. The prepared mixture was maintained at room temperature and placed in a dark environment on a shaker-incubator for 24 hours to prevent any potential photoactivation of ZnCl₂ [20]. During this process, a gradual color change was observed from light yellow to pale white, indicating the formation of ZnO nanoparticles. The pH of the mixture was initially 6.4 and increased to 9.2 after the addition of the plant extract. Following the incubation period, the mixture was centrifuged at 12,000 rpm for 20 minutes. After discarding the supernatant, the remaining pellet was washed with 70% ethanol to eliminate any residual impurities, and centrifugation was repeated. The resulting sediment was then transferred into sterile glass dishes and allowed to air-dry at ambient temperature overnight, resulting in the formation of a dry powder made from zinc oxide nanoparticles. The final yield of ZnO NPs was approximately 0.41 g, corresponding to a synthesis efficiency of 82%.

2.5 Characterization of the plant-based green synthesized zinc oxide nanoparticles

The green synthesized nanoparticles derived from plants were characterized in the laboratory of the Department of Chemistry, University of Baghdad, using the following techniques:

2.5.1 Atomic force microscopy

The surface morphology, granularity, topography, and roughness of the synthesized ZnO-NPs were analyzed using atomic force microscopy (AFM). To prepare the sample for analysis, $100~\mu l$ of the nanoparticle suspension was deposited onto a clean glass slide, forming a thin film. This was then left to dry at room temperature prior to examination with the AFM instrument (AA3000, Angstrom Advanced Inc., USA), operated in tapping mode under ambient conditions [21].

2.5.2 Ultraviolet–visible spectroscopy

To conduct the ultraviolet-visible (UV-Vis) spectroscopy, the absorbance spectra of both the ZnO-NP solution and the

reference were recorded across a wavelength range of 200 to 1100 nm. The resulting spectra were utilized to determine the maximum absorbance (max). The spectral measurements were repeated to ensure accuracy, enabling the estimation of any potential error in the maximum value [22]. A Shimadzu UV-1800 spectrophotometer (Shimadzu Corporation, Japan) was used with a quartz cuvette (1 cm path length). Measurements were taken at room temperature using a deuterium and tungsten-halogen lamp as light sources.

2.5.3 Fourier transform infrared spectroscopy

Fourier transform infrared spectroscopy (FTIR) was used in the Chemistry Department of Al-Nahrain University's College of Science in Iraq to further characterize the ZnO-NPs. To prepare the sample, $100~\mu l$ of the synthesized ZnO-NP solution was subjected to ultrasonic agitation, after which the sample was analyzed at room temperature. The FTIR spectra were recorded within the $400\text{--}4000~\text{cm}^{-1}$ wavenumber range [23]. Spectra were obtained using a Bruker Alpha II FTIR spectrometer equipped with a DTGS detector and a KBr beam splitter. Samples were scanned 32 times at a resolution of 4 cm $^{-1}$.

2.6 Antimicrobial activity of zinc oxide nanoparticles

Xanthomonas citri was subcultured on Muller–Hinton agar plates using the streaking technique with a sterilized inoculating loop, following growth in nutrient broth. Subsequently, five wells were carefully created in each agar plate. Into each well, 50 μL of varying concentrations of ZnO-NPs was introduced, alongside a control (DMSO). The plates were then sealed to prevent contamination and incubated at 37°C for 48 hours. Each treatment was performed in triplicate, and data were collected the following day.

2.7 Statistical analysis

Statistical analysis was conducted using the Statistical Package for the Social Sciences (SPSS; IBM, version 24), following a factorial experimental design. The experiment was arranged in a completely randomized block design (CRBD), with three replicates for each concentration. Additionally, GraphPad Prism (version 10) was employed to evaluate the impact of various concentrations of ZnO-NPs and the plant extract on bacterial growth. Statistical significance between treatment means was determined at $p=0.05\,$

3. RESULTS AND DISCUSSION

3.1 Plant extract sensitivity to Xanthomonas citri

On Muller-Hinton agar, plant extracts at concentrations of S-100, 50, 25, and 12.5 mg/mL were placed in wells, followed by inoculation with an overnight bacterial culture. Upon examining the results the next day, *Xanthomonas citri* was found to be resistant to the plant extract, as illustrated in Figure 1 and Table 1. These findings align with previous research [24]. This study's objective was to assess the plant extract's antimicrobial activity against *Xanthomonas axonopodis* pv. Mangifera indica in a laboratory setting. The disc diffusion method was applied to assess the efficacy of the aqueous extract of *C. erectus*, revealing that the extract was ineffective in inhibiting bacterial growth. The negative control (DMSO)

produced no inhibition zone, confirming that the observed activity (or lack thereof) was due to the test sample itself.

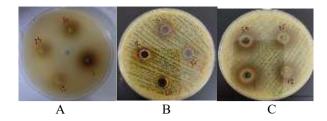


Figure 1. Antibacterial activity of the plant extract against *Xanthomonas citri* at different concentrations (S-100, 50, 25, 12.5 mg/ml)

3.2 Antibacterial efficacy of biosynthesized zinc oxide nanoparticles

The results of the antibacterial efficacy of biosynthesized ZnO-NPs at varying concentrations (100, 50, 25, and 12.5 mg/mL) are presented in Figures 2 and 3 and Table 1. The results unequivocally show a dose-dependent relationship, with higher ZnO-NP concentrations demonstrating stronger antibacterial activity. Table 1 shows that an inhibition zone of 7 mm was observed at a concentration as low as 12.5 mg/mL, whereas an inhibition zone of 18 mm was produced at a concentration of 100 mg/mL. Although no standard antibiotic (e.g., streptomycin) was included in the present assays, previous reports demonstrated that streptomycin sulfate at 500 ppm produced inhibition zones of approximately 17–19 mm against X. axonopodis pv. citri, suggesting that the ZnO-NPs exhibit comparable antimicrobial potential [25]. In comparison with chemically synthesized ZnO-NPs, the biosynthesized nanoparticles in our study demonstrated comparable or even superior antibacterial efficacy. For instance, study [26] reported that chemically synthesized ZnO-NPs produced inhibition zones ranging from 10 to 16 mm against Xanthomonas campestris at concentrations between 50 and 100 mg/mL. In contrast, our green-synthesized ZnO-NPs achieved an inhibition zone of 18 mm at 100 mg/mL against X. citri, suggesting that the phytochemicals present in the plant extract may enhance the antibacterial activity by improving nanoparticle dispersion, stability, and interaction with bacterial membranes [26].

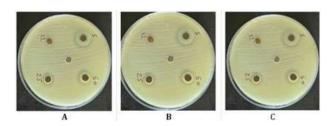


Figure 2. Antibacterial activity of the biosynthesized zinc oxide nanoparticles against *Xanthomonas citri* at varying concentrations (S-100, 50, 25, 12.5 mg/mL)

The variation in inhibition zones, as shown in the table, may be attributed to differences in ZnO-NP's interaction mechanisms with the bacterial cells, as well as the susceptibility of the bacteria used in the study. These observations are consistent with previous research findings [27]. The antibacterial capabilities of ZnO-NPs with varying surface areas against *Xanthomonas campestris* pv. campestris, the efficiency of ZnO-NPs in managing black mold disease, and their impact

on soil bacterial communities were also investigated in that study. The findings revealed that ZnO-NPs with larger surface areas significantly inhibited X. campestris pv. campestris by disrupting bacterial growth and causing cellular damage. The dose-dependent antibacterial activity observed in this study can be explained by the ability of ZnO-NPs to generate reactive oxygen species (ROS), such as hydroxyl radicals, superoxide anions, and hydrogen peroxide, which induce oxidative damage to bacterial cell membranes, proteins, and DNA, ultimately leading to cell death [28, 29]. Additionally, ZnO-NPs interact directly with bacterial membranes, disrupting their integrity and increasing permeability, which causes leakage of cellular contents [30]. These combined mechanisms contribute to the enhanced antimicrobial efficacy observed and are supported by previous literature. The zinc oxide nanoparticles induce bacterial cell death by generating reactive oxygen species (ROS), such as hydroxyl radicals (•OH), superoxide anions (O₂⁻), and hydrogen peroxide (H₂O₂), which cause protein, DNA, and membrane damage. A schematic representation of the proposed antibacterial mechanism is provided in Figure 4.

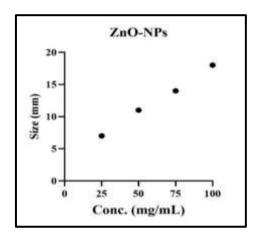


Figure 3. Dose–response curve of biosynthesized zinc oxide nanoparticles against *Xanthomonas citri*

Table 1. Zones of inhibition induced by *Conocarpus* extract and zinc oxide nanoparticles against *Xanthomonas citri* isolates

Test	Concentration	Diameter of Inhibition Zone (mm)		
Solution	(mg/mL)	Isolate 1	Isolate 2	Isolate 3
ZnO-NPs	S	18	20	18
	50	14	13	14
	25	11	12	11
	12.5	7	6	7
	S	Nill	Nill	Nill
Plant	50	Nill	Nill	Nill
extract	25	Nill	Nill	Nill
	12.5	Nill	Nill	Nill

In addition to antimicrobial potency, the long-term stability and practical applicability of ZnO-NPs are crucial for real-world deployment. ZnO-NPs are known for their high chemical stability under diverse conditions, including different pH and environmental stress, and can be further enhanced via surface functionalization [31]. Nonetheless, colloidal aggregation and surface oxidation over time may reduce their activity. Therefore, future studies should assess storage stability over extended periods and under variable temperature, humidity, and light exposure. Additionally, techniques such as polymer coating or nanocarrier

encapsulation could enhance nanoparticle stability and enable controlled release, making them more suitable for agricultural or biomedical field applications.

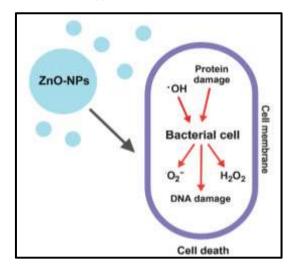
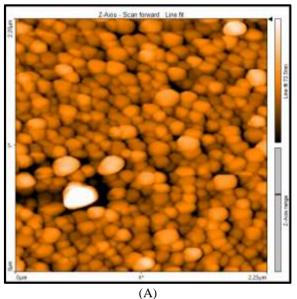
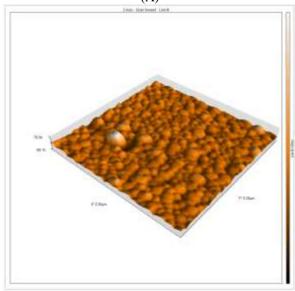


Figure 4. Illustration of the proposed antibacterial mechanism of zinc oxide nanoparticles





(B)

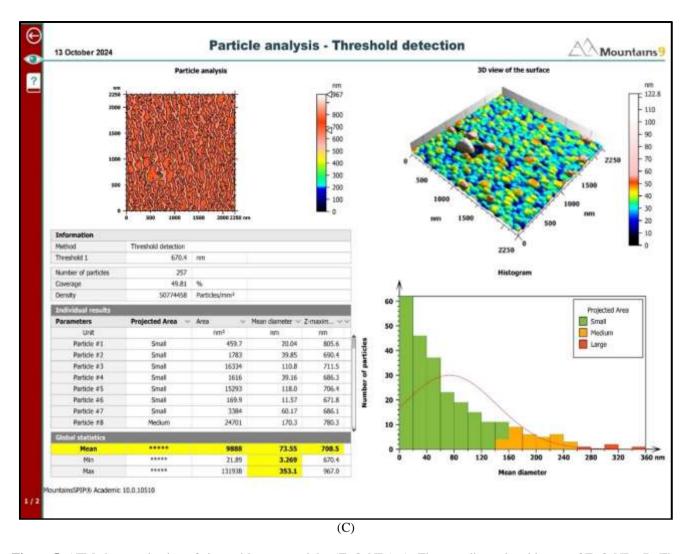


Figure 5. AFM characterization of zinc oxide nanoparticles (ZnO-NPs). A: The two-dimensional image of ZnO-NPs; B: The three-dimensional image of ZnO-NPs; C: AFM analysis report of ZnO-NPs

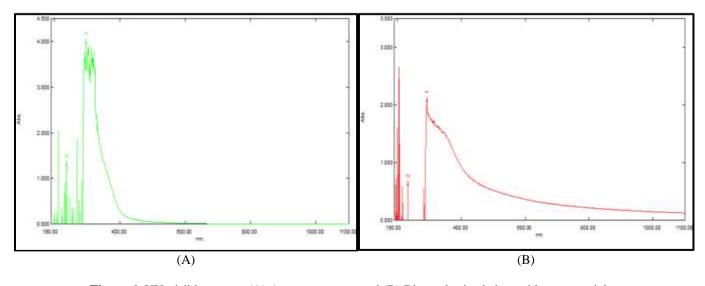


Figure 6. UV-visible spectra (A) Aqueous extract and (B) Biosynthesized zinc oxide nanoparticles

The ZnO-NPs' surface roughness was visible to the AFM. As shown in Figure 5, the results demonstrated that the sample's 2D and 3D images had a uniform height distribution with an average of 73.55 nm. The previous study [32]

examined the AFM and found that pure ZnO-NPs had an average diameter of 88.17 nm. The results showed that ZnO-NPs are pure and effective, making them suitable for use in a range of industrial and medicinal applications. Smaller particle size enhances antibacterial activity due to increased surface area and better interaction with bacterial cells, which likely contributed to the observed inhibition zones.

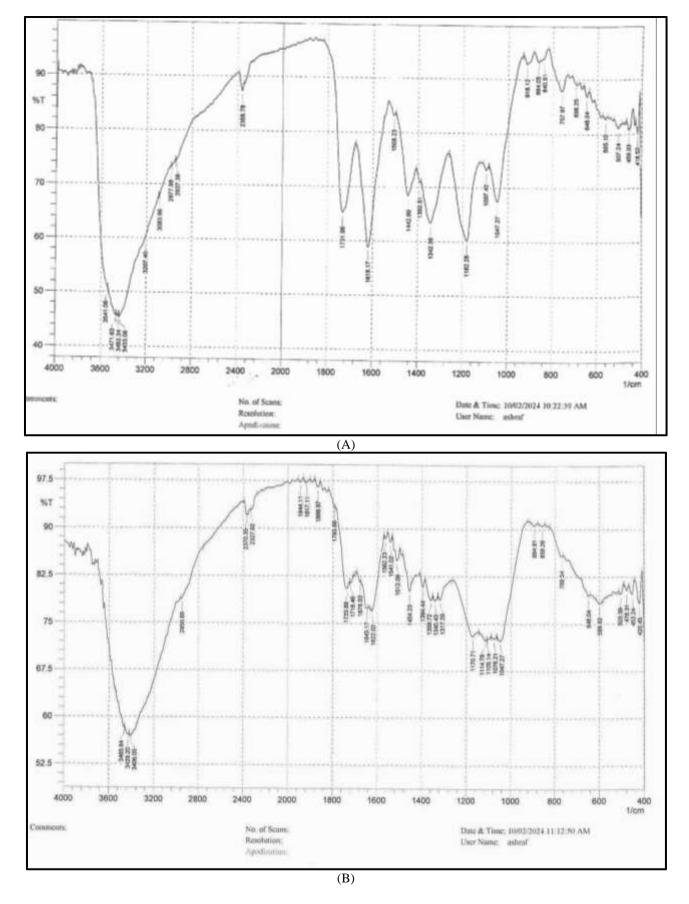


Figure 7. FTIR spectra of (A) Biosynthesized zinc oxide nanoparticles and (B) Plant extract

3.4 UV-Visible absorption of zinc oxide nanoparticles

The results presented in Figure 6(A) demonstrate that the UV-visible spectroscopy of the aqueous plant extract exhibited a characteristic absorption peak at 259 nm. The UV-Vis

spectrum of the green-synthesized ZnO-NPs using the crude extract, on the other hand, is shown in Figure 6(B). It shows an absorption peak at 292 nm. According to previous findings [33], ZnO-NPs typically exhibit a distinct absorption peak near 275 nm, as determined by UV-visible spectroscopy.

3.5 FTIR spectroscopy-based characterization of zinc oxide nanoparticles

The results of the FTIR spectral analysis of the biosynthesized ZnO-NPs are illustrated in Figure 7(A). The spectra display distinct absorption bands at 3471.63 cm⁻¹ and 3433.06 cm⁻¹, which are attributed to O–H stretching vibrations of alcohol groups and N–H stretching from amine groups. Additionally, a band observed at 1618.17 cm⁻¹ corresponds to N–H bending of amines and C=C stretching vibrations. Another notable peak appears at 1182.28 cm⁻¹, which is associated with C–O stretching. Figure 7(B) presents the FTIR spectrum of the plant extract, showing similar characteristic peaks. These spectral patterns are in agreement with findings reported in previous studies [34].

4. CONCLUSIONS

Using the aqueous extract of *C. lancifolius* leaves, this study successfully demonstrated the green synthesis of ZnO-NPs and assessed their antibacterial efficacy against Xanthomonas citri. The biosynthesized ZnO-NPs demonstrated significantly enhanced antimicrobial effects, whereas the crude plant extract showed no discernible antibacterial activity. Characterization of the nanoparticles confirmed their successful formation, with AFM revealing an average particle size of 73.55 nm and UV-visible spectroscopy indicating a distinct absorption peak at approximately 292 nm. Fourier transform infrared spectroscopy further verified the presence of functional groups involved in the nanoparticle formation. The ZnO-NPs displayed notable antibacterial activity, achieving a maximum inhibition zone of 18 mm at a concentration of 100 mg/mL. These findings highlight the potential of green-synthesized ZnO-NPs as an effective biocontrol agent against phytopathogenic bacteria, offering a sustainable and eco-friendly alternative to conventional treatments. A key limitation of this study is that the Conocarpus lancifolius leaf extract alone exhibited no antibacterial activity against Xanthomonas citri. This suggests that the antimicrobial effect observed was solely due to the synthesized zinc oxide nanoparticles. Therefore, the independent antibacterial potential of the plant extract remains unverified, limiting its application without nanoparticle formulation. Moreover, the study was limited to specific Iraqi regions, and in vitro results may not accurately reflect realworld field conditions.

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